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Properties of Ag-Doped $\text{Bi}_{(1.6)}\text{Pb}_{(0.4)}\text{Sr}_2\text{Ca}_3\text{Cu}_{(4-x)}\text{Ag}_x\text{O}_y$ (2234) Oxides Prepared by S.S.R. Method

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Abstract: The effect of Ag-doping $\text{Bi}_{1.6}\text{Pb}_{0.4}\text{Sr}_2\text{Ca}_3\text{Cu}_{4-x}\text{Ag}_x\text{O}_y$ compounds ($x=0.0-1.0$), prepared by conventional Solid-State-Reaction (SSR) technique, was studied using x-ray diffraction (XRD) and electrical resistivity. The high- T_c fraction of the 2223 phase, formed from the nominal composition of 2234, decreases with increasing Ag content. From lattice parameter calculations it follows that Ag-doping the unit cell phase of $(\text{Bi,Pb})_2\text{Sr}_2\text{Ca}_2\text{Cu}_3\text{O}_y$ is limited to the value of $x \leq 0.1$. The zero resistance critical temperature ($T_{c,\text{zero}}$) was determined from the resistivity curves for all samples. The value of $T_{c,\text{zero}}$ decreases slightly to 106 K for $x \leq 0.4$ and when the nominal silver content increases up to $x=0.7$ or more $T_{c,\text{zero}}$ strongly decreases to 72 K. The critical current density J_c at 77 K decreases dramatically as silver content increases. This result is discussed on the basis of precipitation of low- T_c 2212 and other impurity phases in $\text{Bi}_{1.6}\text{Pb}_{0.4}\text{Sr}_2\text{Ca}_3\text{Cu}_{4-x}\text{Ag}_x\text{O}_y$ compounds.

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