

Faculty Profile

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Education and Appointments

- B.A. 1963, New York University
- Ph.D. 1967, Harvard University

Honors

- Fellow of the American Association for the Advancement of Science, 1984
- Faculty of Science Career Teaching Award, 1990

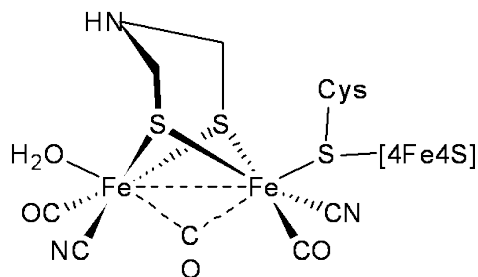
Research Interests

- Organic
- Bioinorganic
- Bioorganic
- Energy Science

Research Summary

Mechanistic, Synthetic and Structural Chemistry

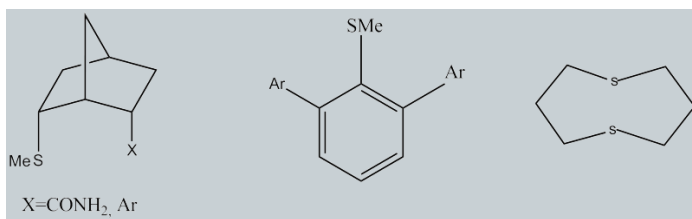
H₂ is an ideal storable form of energy. Splitting water produces H₂ and burning it, or even better using it in fuel cells, regenerates water and releases the chemical energy stored in H₂. Thus H₂ can serve as a recyclable alternative energy source which avoids CO₂, a major contributor to global warming. Electrolysis of water to produce H₂ and H₂ fuel cells require catalysis by Pt, a rare and expensive metal. However, [Fe Fe]-hydrogenase, whose active site is shown below,



is an exceptionally efficient enzyme (high turnovers rates at low overpotentials) for producing H₂ and uses cheap and readily available iron. Inspired by Nature we have synthesized organometallic catalysts with Fe₂S₂ cores that mimic the enzyme active site to convert transient electricity and solar energy into the storable chemical energy inherent in H₂ in collaboration with Professors Lichtenberger and Evans (Purdue). Understanding the mechanism for catalysis through

experimental and computational methods, is enabling us to design more effective synthetic catalysts.

Electron transfer is of fundamental importance in such areas as biological redox reactions, organometals, superconductors, and photoconductors. We are interested in the factors that control electron transfer. Since sulfur is cheap, implicated in biological redox chemistry and sulfur compounds are useful in organometals, superconductors and photoconductors, it is a featured atom in our research. We have found that juxtapositioning thioethers and electron rich groups close to each other in space with the correct geometry may result in exceptional redox chemistry. To accomplish these juxtapositions conformationally constrained systems must be synthesized. Three such molecularly constrained systems, which show unusually facile oxidation are listed below.



Selenium is an essential mineral and is required in small amounts in the diet of humans. However, larger amounts are toxic. A number of mammalian and bacterial proteins containing selenium, have been found and typically the selenium occurs as selenocysteine and, if the protein is an enzyme, the selenocysteine is at the active site. Selenocysteine has been found to be the 21st amino acid incorporated contranationally. We have shown that the biosynthesis of selenocysteine involves the intermediacy of the novel monoselenophosphate which is the biological selenium donor formed from ingested selenite. Identification of biological Se-containing species by Se-77 NMR spectroscopy is a powerful methodology which we are developing.

Selected Publications

Catalysts for H₂ Generation

- Felton, G.A.N.; Vannucci, A.K.; Chen, J.; Lockett, L.T.; Okumura, N.; Petro, B.J.; Zakai, U.I.; Evans, D.H.; Glass, R.S.; Lichtenberger, D.L. "Hydrogen Generation from Weak Acids: Electrochemical and Computational Studies of a Diiron Hydrogenase Mimic," *J. Am. Chem. Soc.* 2007, *129*, 12521-12530.
- Felton, G.A.N.; Petro, B.J.; Glass, R.S.; Lichtenberger, D.L.; Evans, D.H.; "One to Two Electron Reduction of an [FeFe]-Hydrogenase Active site Mimic: The Critical Role of Fluxionality of the [2Fe2S] Core," *J. Am. Chem. Soc.* 2009, *113*, 11290-11291.
- Chen, J.; Vannucci, A.K.; Mebi, C.A.; Okumura, N.; Borowski, S.C.; Swenson, M.; Lockett, L.T.; Evans, D.H.; Glass, R.S.; Lichtenberger, D.L. "Synthesis of Diiron Hydrogenase Mimics Bearing Hydroquinone and Related Ligands. Electrochemical and Computational Studies of the Mechanism of Hydrogen Production and the Role of O-H • • • S Hydrogen Bonding," *Organometallics* 2010, *29*, 5330-5340.

Electron Transfer

- Glass, R.S.; Hug, G.L.; Schöneich, C.; Wilson, G.S.; Kuznetsova, L.; Lee, T.-M.; Ammam, M.; Lorange, E.; Nauser, T.; Nichol, G.S.; Yamamoto, T. "Neighboring Amide Participation in Thioether Oxidation: Relevance to Biological Oxidation," *J. Am. Chem. Soc.* 2009, *131*, 13791-13805.
- Chung, W.J.; Ammam, M.; Gruhn, N.E.; Nichol, G.S.; Singh, W.P.; Wilson, G.S.; Glass, R.S. "Interactions of Arenes and Thioethers Results in Facilitated Oxidation," *Org. Lett.* 2009, *11*, 397-400.
- Ammam, M.; Zakai, U.I.; Wilson, G.S.; Glass, R.S. "Anodic Oxidation of m-Terphenylthio-seleno- and telluroethers: Lowered Oxidation Potentials Due to Chalcogen • • • π Interactions." *Pure Appl. Chem.* 2010, *82*, 555-563.
- Evans, D.H.; Gruhn, N.E.; Jin, J.; Li, B.; Lorange, E.; Okumura, N.; Macias-Ruvalcaba; Zakai, U.I.; Zhang, S.-Z.; Block, E.; Glass, R.S. "Electrochemical and Chemical Oxidation of Dithia-, Ditellura-, Selenathia-, and Tellurathia-Mesocycles and Stability of the Oxidized Species," *J. Org. Chem.* 2010, *75*, 1997-2009.

- Glass, R.S.; Schöneich, C.; Wilson, G.S.; Nauser, T.; Yamamoto, T.; Lorange, E.; Nichol, G.S.; Ammam, M. "Neighboring Pyrrolidine Amide Participation in Thioether Oxidation. Methionine as a Hopping Site," *Org. Lett.* 2011, 13, 2837-2839.

Chemical Biology of Selenium

- Xu, X.-M.; Carlson, B.A.; Mix, H.; Zhang, Y.; Saira, K.; Glass, R.S.; Berry, M.J.; Gladyshev, V.N.; Hatfield, D.L. "Biosynthesis of Selenocysteine on its tRNA in Eukaryotes," *PLoS Biol.*, 2007, 5, 96-105.
- Glass, R.S.; Berry, M.J.; Block, E.; Boakye, H.T.; Carlson, B.A.; Gailer, J.; George, G.N.; Gladyshev, V.N.; Hatfield, D.L.; Jacobsen, N.E.; Johnson, S.; Kahakachchi, C.; Kaminski, R.; Manley, S.A.; Mix, H.; Pickering, I.J.; Prenner, E.J.; Saira, K.; Skowronksa, A.; Tyson, J.F.; Uden, P.C.; Wu, Q.; Xu, X.-M.; Yamdagni, R.; Zhang, Y. "Insights into the Chemical Biology of Selenium," *Phosphorus, Sulfur, Silicon and the Related Elements* 2008, 183, 924-930.
- Glass, R.S. "Redox Chemistry of Sulfur, Selenium and Tellurium Compounds," in *Selenium and Tellurium Chemistry*, Wollins, J.D., Laitinen, R.S. Eds., Springer-Verlag, Berlin, 2011, pp 57-77.

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