



TOP > Available Issues > Table of Contents > Abstract

ONLINE ISSN : 1348-2246 PRINT ISSN: 0910-6340

Analytical Sciences Vol. 26 (2010), No. 7 p.791

[PDF (477K)] [References]

Analytical Parameters for Amplitude-Modulated Multiplexed Flow Analysis

Yohei KUROKAWA¹⁾, Masaki TAKEUCHI²⁾ and Hideji TANAKA²⁾

1) Graduate School of Pharmaceutical Sciences, Tokushima University 2) Institute of Health Biosciences, Tokushima University

(Received April 8, 2010) (Accepted May 6, 2010)

Analytical conditions of amplitude-modulated multiplexed flow analysis, the basic concept of which was recently proposed by our group, are investigated for higher sample throughput rate. The performance of the improved system is evaluated by applying it to the determination of chloride ions. The flow rates of two sample solutions are independently varied in accordance with sinusoidal voltage signals, each having different frequency. The solutions are merged with a reagent solution and/or a diluent, while the total flow rate is held constant. Downstream, the analytical signal V_{d} is monitored with a spectrophotometer. The

 $V_{\rm d}$ shows a complicated profile resulting from amplitude modulated and multiplexed information on the two samples. The V_{d} can, however, be deconvoluted to the contribution of each sample through fast Fourier transform (FFT). The amplitudes of the separated wave components are closely related to the concentrations of the analytes in the samples. By moving the window for FFT analysis with time, a temporal profile of the amplitudes can be obtained in real-time. Analytical conditions such as modulation period and system configuration have been optimized using aqueous solutions of Malachite Green (MG). Adequate amplitudes are obtained at the period of as low as 5 s. At this period, the calibration curve for the MG concentration of $0 - 30 \ \mu mol \ dm^{-3}$ has enough linearity ($r^2 =$ 0.999) and the limit of detection (3.3σ) is 1.3 µmol dm⁻³; the relative standard deviation of repeated measurements ($C_{MG} = 15 \ \mu \text{mol dm}^{-3}$, n = 10) is 2.4%. The developed system has been applied to the determination of chloride ions by a mercury(II) thiocyanate method. The system can adequately follow the changes in analyte concentration. The recoveries of

chloride ion spiked in real water samples (river and tap water) are satisfactory, around 100%.

[PDF (477K)] [References]

Download Meta of Article[<u>Help</u>] <u>RIS</u> <u>BibTeX</u>

To cite this article:

Yohei KUROKAWA, Masaki TAKEUCHI and Hideji TANAKA, *Anal. Sci.*, Vol. 26, p.791, (2010).

doi:10.2116/analsci.26.791 JOI JST.JSTAGE/analsci/26.791

Copyright (c) 2010 by The Japan Society for Analytical Chemistry

