研究论文

丁二酮肟双核铜配合物与DNA相互作用的电化学研究

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摘要 用电化学方法研究了丁二酮肟双核铜配合物[Cu2(Hdmg),]与DNA的相互作用. 考察了pH、温度、离子 强度和配合物浓度等因素对配合物与DNA相互作用的影响,初步探讨了配合物与DNA相互作用的机理. 实验结果 表明, 配合物与DNA的碱基结合形成非电活性物质, 使溶液中游离配合物的浓度降低, 配合物的峰电流减小. 单链 DNA(ssDNA)充分暴露的碱基使其与配合物的结合能力大于双链DNA(dsDNA). Cu₂(Hdmg)。与ssDNA和 dsDNA的结合比分别为2:1和1:1,结合常数分别为3.56× 10^9 和2.75× 10^5 .

关键词 双核铜配合物 丁二酮肟 DNA 循环伏安法 相互作用

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Electrochemical Studies on Interaction Between Dinuclear Copper(II) Complex Containing Dimethylglyoxime and DNA

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Abstract Cyclic voltammetry was used to study the interaction between dinuclear copper(II) complex containing dimethylglyoxime[Cu2 (Hdmg)4] and DNA. The influences of pH, temperature, ionic strength and concentration of $Cu_2(Hdmg)_4$ on the interaction were described, and the mechanism of the interaction was also discussed. The result shows that $Cu_2(Hdmg)_4$ can interact with the bases of DNA to form non-electroactive complex, which can decrease the anodic peak current of $Cu_2(Hdmg)_4$. The interaction between $Cu_2(Hdmg)_4$ and ssDNA is stranger than that between Cu₂(Hdmg)₄ and dsDNA because of the exposed bases of ssDNA. The combining ratios of $Cu_2(Hdmg)_4$ to ssDNA and to dsDNA are 2:1 and 1:1, and the combining constants 3.56×10^9 and 2.75×10^5 , respectively.

Key words Dinuclear copper(II) complex; Dimethylglyoxime; DNA; Cyclic voltammetry; Interaction

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