芳基-8-喹啉氧基汞化合物的合成与波谱性质研究

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摘要 本文合成了一系列共13种芳基-8-喹啉氧基汞化合物,其中12种为新化合物,经IR,^1H NMR和元素分析鉴定了其组成与结构,紫外光谱研究表明分子中存在有N→Hg分子内配位. 提出了溶剂效应对紫外光谱性质和N→Hg分子内配位影响的机理. 核磁共振氢谱表明取代基仅影响汞-苯环质子的化学位移,说明共轭效应未通过分子内配位贯穿于整个分子.^1^9^9Hg化学位移值与Hammett-Brown常数σ^+间存在有良好的线性关系

关键词 <u>红外分光光度法</u> <u>元素分析</u> <u>质子磁共振谱法</u> <u>溶剂效应</u> <u>化学位移</u> <u>羟基喹啉类</u> 河南省自然科学基金 分子内配体 <u>芳基喹啉氧基汞</u> 氯化芳基汞

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# Synthesis and spectral properties of arylmercuric quinolin-8-olate

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Abstract Thirteen arylmercuric quinolin-8-olates were synthesized and characterized by IR, 1H NMR and elemental anal., of which twelve are new compounds UV spectra showed that an intramol. N ?Hg coordination exists in the mols. examined The mechanism of the solvent effect on the UV spectral properties and on the N ?Hg coordination was proposed. 1H NMR spectra indicated that the substituents influence only the chem. shifts of the Hg-Ph protons, which reveal that the conjugation does not spread throughout the whole mols. via the intramol. coordination. A good linear relation was found between the 199Hg chem. shifts and the Hammett-Brown s+ constants

Key wordsINFRARED SPECTROPHOTOMETRYELEMENTAL ANALYSISPROTON MAGNETICRESONANCE SPECTROMETRYSOLVENT EFFECTCHEMICAL SHIFTHYDROXY QUINOLINES

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