

研究报告

## Fe<sub>3</sub>O<sub>4</sub>纳米磁性微粒对钴和锶的吸附

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### 摘要

为降低90Sr和60Co对环境的污染,用共沉淀法制备了粒径为10 nm的Fe<sub>3</sub>O<sub>4</sub>磁性微粒,分散于水中生成饱和磁化强度(M)为350 kA/m的水基磁流体,用此磁流体对Co<sup>2+</sup>, Sr<sup>2+</sup>进行了吸附研究。结果表明,在4 °C,吸附60 min时,Co<sup>2+</sup>, Sr<sup>2+</sup>分别在pH=7和pH=8下达到吸附平衡,吸附容量为1.794, 0.962 mmol/g。用Langmuir等温模型、假二级动力学模型探讨了Fe<sub>3</sub>O<sub>4</sub>纳米磁性微粒对Co<sup>2+</sup>, Sr<sup>2+</sup>的吸附机制,研究结果表明,该过程是单离子层吸附过程。

关键词 钴 锶 吸附 Fe<sub>3</sub>O<sub>4</sub>纳米磁性微粒

分类号 0647.3

## Adsorption of Cobalt and Strontium on Fe<sub>3</sub>O<sub>4</sub> Magnetic N anoparticles

### Abstract

Adsorption is an efficiency method to decrease the pollution of 90 Sr and 60 Co. The adsorption behavior of cobalt and strontium on Fe<sub>3</sub>O<sub>4</sub> magnetic nanoparticles was studied in this paper. The adsorbent applied here was prepared by co-precipitation, and can form water based ferrofluids. Its saturation magnetization is 350 kA/m. The experiments show that the  $Q_{\max}$  of cobalt and strontium are 1.794 mmol/g and 0.962 mmol/g at pH=7 and pH=8, respectively. The adsorption process follows a pseudo second kinetics and the Langmuir isotherm, which indicates the adsorption is a monolayer adsorption.

### Key words

cobalt strontium adsorption Fe<sub>3</sub>O<sub>4</sub> magnetic nanoparticles

### 扩展功能

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