葡萄糖和水混合溶液中多组份电解质溶液热力学I: HCl-NaCl-D-glucose-H2O体系(5-45℃)

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摘要 本文在恒定葡萄糖质量百分数x=10%的条件下,应用电动势法测定无液体接界电池(A)和电池(B)的电动势: Pt,H2(1.013X10^5 Pa)|HCl(m),D-Glucose(x),H2O(1-x)|AgCl-Ag Pt,H2(1.013X10^5 Pa)|HCl(mA),NaCl(mA),D-Glucose(x),H2O(1-x)|AgCl-Ag (B) 根据电池(A)电动势确定混合液中的Ag-AgCl电极的标准电极电势,讨论了HCl的迁移性质;利用电池(B)的电动势确定了HCl在该体系中的活度系数γA,结果表明,在恒定总离子强度下,HCl的活度系数服从Harned 规则。在溶液组成恒定时,lgγA是温度倒数1/T的线性函数,进一步讨论了混合物中HCl的相对偏摩尔焓,计算了HCl的介质效应。

 关键词
 水
 热力学性质
 盐酸
 葡萄糖
 电动势
 氯化钠
 焓
 混合液
 电解液

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# Thermodynamics of mixed electrolyte in mixed solutios of D-glucose and water. I:system of HCl-NaCl-D-Glucose-H2O from 5 to 45

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**Abstract** The transfer properties of HCl were calculated and the activity coefficients of HCl in the system were determine The activity of HCl obeyes Harned up to I = 2.0 mol/kg over a range 5-45? Under the condition of constant components of mixture, the logarithm of HCl activity coefficients are a linear function with reciprocal of temps. The primary, secondary and total medium effect of HCl were calculated and discussed.

 Key words
 WATER
 THERMODYNAMIC PROPERTIES
 HYDROCHLORIC ACID
 GLUCOSE

 ELECTROMOTIVE FORCES
 SODIUM CHLORIDE
 ENTHALPY
 MIXED SOLTION
 ELECTROLYTES

DOI:

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