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论文

苯氧乙酸嗜霉胺盐的低温热容和热力学性质研究

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摘要:

用精密自动绝热量热计测定了苯氧乙酸嗜霉胺盐在81-380 K之间的低温热容。结果表明, 该化合物在81-328 K之间无相变和热异常现象发生, 在328-354 K之间发生固-液熔化, 其熔化温度、摩尔熔化焓和摩尔熔化熵分别为 (349.38 ± 0.03) K, (34.279 ± 10) kJ/mol和 (98.13 ± 0.05) J/(K·mol)。根据热力学函数关系式计算出苯氧乙酸嗜霉胺盐在80-325 K之间以标准状态(298.15 K)为基准的热力学函数值。

关键词: 苯氧乙酸嗜霉胺盐 热容 热力学函数 绝热量热法

Heat Capacity and Thermodynamic Properties of Crystalline Pyrimethanil Phenoxyacetate($C_{20}H_{21}N_3O_3$)SUN Xiao-Hong^{1*}; LIU Yuan-Fa²; TAN Zhi-Cheng³; JIA Ying-Qi²; WANG Mei-Han³

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Abstract:

Low-temperature heat capacities of Pyrimethanil phenoxyacetate($C_{20}H_{21}N_3O_3$) were precisely measured with a high-precision automated adiabatic calorimeter over the temperature range between 81 to 380 K. The compound was observed to melt at (349.38 ± 0.04) K. The molar enthalpy $\Delta_{fus}H_m$, and entropy of fusion, $\Delta_{fus}S_m$ of the compound were determined to be (34.279 ± 10) J/mol, (98.13 ± 0.05) J/(mol·K), respectively. The thermodynamic function data relative to the reference temperature (298.15 K) were calculated based on the heat capacities measurements in the temperature range from 80 K to 325 K.

Keywords: Pyrimethanil phenoxyacetate Heat capacity Thermodynamic function Adiabatic calorimetry

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