$CH(A^2 \triangle \pi B^2 \sum^-)$ 和醇类分子碰撞过程动力学研究

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摘要 本文报道用四倍频YAG激光(266nm)光解CHBr3产生电子激发态CH(A,B)自由基和测量自发辐射CH (A,B→X)的时间分辨信号的方法测定了室温(290K)下CH(A,B)被醇类分子(乙醇、异丙醇、正丁醇、异戊醇和叔戊醇)猝灭的速率常数,实验测定的CH(A)和CH(B)猝灭速率常数k~q^A和k~q^B(单位为10^-1^0cm^3.molecule^-^1.s^-^1)值如下(误差为线性拟合的标准偏差):此外,还从碰撞配合物模型出发,就醇分子中OH基对猝灭速率常数的影响作了讨论。

关键词 醇 动力学研究 猝灭 碰撞

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Kinetic study of CH(A,B) quenching process by alcohol molecules

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Abstract The quenching rate constants of CH(A and B) by alcohol Molecules were measured by using laser photolysis (at 266nm) of CHBr3 to produce CH(A,B) and the observation of time resolved fluorescence CH(A,B \rightarrow X) methods. The measured results are as follows (units of kq in $10^{-1}^{0.4}$ molecule $^{-1.5}$; the uncertainty is a standard deviation of least-squares fitting): A discussion of the effect of OH group in alcohol on the quenching of CH(A,B) is presented based on the collision complex model.

Key words ALCOHOL KINETIC STUDY QUENCHING COLLISON

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