微乳液相行物为结构的研究:氯化钠/水/十二烷基硫酸钠/正丁醇/正辛烷系统

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摘要 本文报道NaCl/H2O/SDS/n-C4H90H/n-C8H18系统在WC4H9OH/WSDS=4时,不同盐浓度下的相图,结果表明,卤水中NaCl浓度的增大导致这个系统相分裂区域扩在,

并在一定范围内产生三相共存和出现中相微乳液。中相微乳液的密度,粘度,

电导率和折光率与最佳盐浓度以及微观物型转变有对应关系,红外光谱测定表明,

油相中氢键的缔合强度比中相微乳液显著增强。

关键词 <u>红外分光光度法</u><u>水</u><u>氯化钠</u><u>缔合</u><u>氢键</u><u>硫酸钠</u><u>十二碳化合物</u><u>相结构</u><u>相图</u><u>微观结构</u> 正辛烷 正丁醇

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Phase behavior and micro-structure of microemulsions: Nacl/H2o/SDs/u/C4H9OH/n-C8H18 system

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Abstract Phase diagrams were constructed for the NaCl/H2O/SDS/BuOH/n-C8H18 system at BuOH:SDS wt. ratio 4 at various salinities. The diagrams indicate that an increase in salinity leads to expansion of the phase-sepn. region and the formation of a 3-phase area including a middle microemulsion phase. The d., viscosity, elec. conductivity, and n of the middle microemulsion phase have a close relation to the optimal salinity and the transition of microemulsion microstructure. The IR measurements indicate that the H bond strength in the oil phase is notably stronger than that in the middle microemulsion phase.

Key wordsINFRARED SPECTROPHOTOMETRYWATERSODIUM CHLORIDEASSOCIATIONHYDROGEN BONDSSODIUM SULFATEC12 COMPOUNDSPHASE STRUCTUREPHASE DIAGRAMMICRO-STRUCTURENORMAL OCTANEN-BUTANOL

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