研究论文

 $\mathbf{R_2C} = \mathbf{GeH_2}$ 和 $\mathbf{R_2Ge} = \mathbf{CH_2}$ 结构与成键特征的理论研究

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摘要 用密度泛函理论(DFT), 在B3LYP/6-31+G(d, p)水平上研究了取代基对二取代锗烯 R_2 Ge= CH_2

和R₂C=GeH₂ [R=H, OH, NH₂, SH, PH₂, F, Cl, Br, (NHCH)₂, CH₃, (CH)₂]的影响.

研究发现π供电子取代基在碳上时更能引起分子结构在锗端的锥型化.

碳原子上的 π 电子给予取代基的给电子效应越强, R_2 C的单-三态能量差越大, π 电子的反极化效应就越强, 使得化合物的结构在锗端发生的弯曲越明显, 从而使得弯曲结构更稳定. 和前人的计算相比,

碳上的给电子取代基对GeH,结构影响大于它对SiH,的影响.

关键词 密度泛函理论 锗烯 反极化效应

分类号

Theoretical Study on the Structure and Bonding Character of the R₂Ge=CH₂ and R₂C=GeH₂

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Abstract Substituent effects on the disubstituted heavier analogues of $R_2C=GeH_2$ and $R_2Ge=CH_2$ [R=H, OH, NH₂, SH, PH₂, F, Cl, Br, (NHCH)₂, CH₃, (CH)₂] were investigated using density functional theory [B3LYP/6-31+G(d, p)]. The major conclusion was that the stronger the π-donation of the substituents, the larger the ΔE_{ST} of CR₂ and the stronger the π-electron reversed-polarization effect. As the π-electron reversed-polarization effect increased, the distortion of GeH₂ fragment increased, and the stability of the bent structure increased. In addition, the π-electron-donating substitution at the carbon atom was more effective at inducing a bent structure than at the Ge atom. In contrast to the $R_2C=SiH_2$ system, the π-electron-donating substituents at the carbon atom have more effective influence on inducing GeH₂ a bent structure than that of SiH₂.

Key words density functional theory (DFT) germene reversed polarization effect

DOI:

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