

ANTA的结构、性质及其互变异构的理论研究

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摘要 对3-硝基-5-氨基-1,2,4-三唑(ANTA)的三种异构体, 1H-ANTA(I),2-ANTA(II)和4H-ANTA(III)在, binitio-HF/3-21G和DFT-B3LYP/3-21G势能面计算的基础上, 进行6-311G^{***}几何参数全优化, MP2

总能量和SCRF溶剂(四氢呋喃)效应计算。以振动分析和统计热力学为基础,

作标题物热力学性质以及 I 和 II 之间的互变异构反应计算, 求得分子几何, 电子结构和300~1000K范围的焓、熵和热容以及 I 和 II 互变异构平衡常数和速率常数。发现在三种异构体中在通常温度下以 II 在气相下最稳定, I 在溶液中最稳定。低温下难以发生异构化反应, 温度可提高 I 与 II 之间的互变速率, 在

800K时两种异构体在气相中等量共存; 大于800K时 I 更为稳定。

关键词 [动力学](#) [炸药](#) [硝基](#) [氨基](#) [三唑P](#) [从头计算法](#) [互变异构](#) [热力学性质](#) [稳定性](#) [ANTA](#)

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Theoretical study on structures and properties of ANTA and its tautomerization

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Abstract The fully optimized molecular geometries and electronic structures of isomers of 5-amino-3-nitrotriazole [1H-ANTA(I), 2- ANTA(II) and 4H-ANTA(III)] are obtained by using ab initio method at HF/3-21G level and DFT method at B3LYP/ 6-311G^{***} level based on the potential energies surface scanning at the HF/3-21G and B3LYP/3-21G level, respectively. The accurate energies at MP2/6-311G^{***}/HF/ 6- 311G^{***}level and solavtion effects (in tetrahydrofuran) of three isomers have also been obtained. On the bases of the vibrational analysis and statistical thermodynamic theory, the thermodynamic parameters (enthalpies, entropies and heat capacities) for titled compounds, the equilibrium constants and rate constants of tautomerization of I \leftrightarrow II from 300K to 1000K, were calculated. The results show that II is the most stable among three titled compounds in gaseous state at ordinary temperature, and I is the most stable in liquid phase. The tautomerization reaction I \leftrightarrow II is difficult to occur at lower temperature. The tautomerism rate of 1H- ANTA and 2H-ANTA is up when the temperature increasing. II and I are equally coexisted in contents at 800K. I is more stable than II at above 800K.

Key words [DYNAMICS](#) [EXPLOSIVES](#) [NITRO GROUP](#) [AMINO GROUP](#) [PYRRODIAZOLE P](#) [AB INITIO CALCULATION](#) [TAUTOMERISM](#) [THERMODYNAMIC PROPERTIES](#) [STABILITY](#)

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