

C₄₋₀, C₄₋₀⁺, Nb@C₄₋₀⁺, NbC₃₋₉⁺, Nb@C_{4-0H-4}⁺的量子化学研究

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摘要

用量子化学从头计算方法研究了C₄₋₀, C₄₋₀⁺, Nb@C₄₋₀⁺, NbC₃₋₉⁺, Nb@C_{4-0H-4}⁺的几何构型、电子结构和C₂₋₈一样, C₄₋₀(T-d)基态也为⁵A₂态, 笼骨架上具有四个悬挂键。计算结果表明C₄₋₀和C₄₋₀⁺比NbC₃₋₉⁺和Nb@C₄₋₀⁺稳定, 与实验结果一致。

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Quantum chemical study of C₄₋₀, C₄₋₀⁺, Nb@C₄₋₀⁺, NbC₃₋₉⁺, Nb@C_{4-0H-4}⁺

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Abstract Ab initio Hartree-Fock calculations were performed on the equilibrium geometries and electronic structures of a series of endohedral, exohedral and endohedral-exohedral complexes of C₄₋₀. The C₄₋₀(T-d) cage is found to have four unpaired electrons with a ⁵A₂ open-shell ground state and have four dangling bonds. C₄₋₀(T-d) behaves as a sort of hollow superatom with an effective valence of 4, both toward the outside and inside of the carbon cage, so it is possible to form the endohedral metallofullerene Nb@C₄₋₀, exohedral complex C_{4-0H-4} and endohedral-exohedral complex Nb@C_{4-0H-4} from C₄₋₀. From the values of binding energies per atom, it's found that C₄₋₀(T-d) is more stable than C₄₋₀(C_{3-v}), while C₄₋₀⁺(C_{3-v}) is more stable than C₄₋₀⁺(T-d). In networked metallofullerenes NbC₃₋₉⁺, Nb is connected directly with three carbon atom, forming three Nb-C single bonds with the Nb atom protruding from the surface of the carbon cage. Our calculated results show that C₄₋₀ and C₄₋₀⁺ are more stable than NbC₃₋₉⁺, and NbC₃₋₉⁺ is more stable than Nb@C₄₋₀⁺. The results are consistent with the experimental results. Through the comparison of the C₄₋₀ series clusters with the C₂₋₈ and related compounds, we have found that there are many similarities between C₄₋₀ and C₂₋₈. Our calculated results may shed light on other endohedral and exohedral complexes of fullerenes and networked type metallofullerenes in general.

Key words [NIOBIUM COMPLEX](#) [FULLERENES](#) [CAGE STRUCTURE](#) [ELECTRONIC STRUCTURE CONFIGURATION](#) [AB INITIO CALCULATION](#)

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