

纳米TiO₂膜修饰电极异相电催化还原马来酸

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摘要 通过电化学合成先驱体和溶胶-凝胶法在Ti表面修饰一层纳米TiO₂膜, SEM, XRD测试表明晶型为锐钛矿型, 晶粒平均尺寸为25 nm。采用循环伏安法、循环方波伏安法和电解合成法研究了纳米TiO₂膜电极在硫酸介质中的氧化还原行为以及对马来酸(maleic acid)还原的电催化活性。结果表明, 纳米TiO₂膜电极在阴极扫描时有两对可逆氧化还原峰, 可逆半波电位E_(1/2)~r分别为-0.53 V和-0.92 V (vs. SCE, 扫描速度0.05 V·s⁻¹), 对应于TiO₂/Ti₂O₃和TiO₂/Ti(OH)₃两个氧化还原电对的可逆电极过程。其中TiO₂/Ti₂O₃电对对马来酸具有异相电催化还原活性, 纳米TiO₂膜中的Ti~(IV)/Ti~(III)氧化还原电对作为媒质间接电还原马来酸为丁二酸(butane diacid), 反应机理为电化学偶联随后化学催化反应(EC')机理。

关键词 [纳米材料](#) [二氧化钛](#) [电催化](#) [循环伏安法](#) [马来酸](#) [丁二酸](#) [扫描电子显微镜](#) [X射线衍射分析](#)

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Heterogeneous Electrocatalytic Reduction of Maleic Acid on Nanocrystalline TiO₂ Film Modified Ti Electrode

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Abstract The precursor Ti(OEt)₄ was prepared by anodic dissolution of metallic titanium in absolute ethanol and direct hydrolysis to prepare nanocrystalline TiO₂ film on titanium electrode (Ti/nano-TiO₂) by a sol-gel process. SEM and XRD were used to characterize the structure of nanocrystalline TiO₂ film (anatase, 25 nm). Redox behavior and electrocatalytic activities of the Ti/nano-TiO₂ electrode were investigated by cyclic voltammetry (CV) and cyclic voltammetry (COV) and bulk electrolysis. The results indicate that there are two pairs of well-defined redox peaks for Ti/nano-TiO₂ film electrode in 1 mol/L H₂SO₄ with E_(1/2)~r of -0.53 V and -0.92 V (vs. SCE) at 0.05 V·s⁻¹ in correspondence with TiO₂/Ti₂O₃ and TiO₂/Ti(OH)₃ reversible electrode process and the heterogeneous electrocatalytic reduction activities of TiO₂/Ti₂O₃ redox in the electrode for maleic acid. It is found that the indirect electroreduction of maleic acid to butane diacid was achieved by Ti~(IV)/Ti~(III) redox system on the nanocrystalline TiO₂ film surface, the electrode reaction mechanism is the called catalytic (EC) mechanism.

Key words [NANOPHASE MATERIALS](#) [TITANIUM DIOXIDE](#) [ELECTRO-CATALYSIS](#) [CYCLOVOLTAMGRAPH](#) [maleic acid](#) [butane diacid](#) [SEM](#) [XRD](#)

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