Full Papers

Pt-MoO_v/CNT电极的制备及其对乙醇氧化电催化性能研究

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摘要 以 MoO_x 代替贵金属Ru作为辅助催化剂,采用电位阶跃法在 MoO_x /碳纳米管(CNT)电极上电化学沉积Pt纳米颗粒,制备了Pt-MoOx/CNT电极。采用扫描电子显微镜、

X射线能量色散谱仪和循环伏安法对所制备的 $Pt/MoO_x/CNT$ 电极进行了表征,

考察了Pt/MoO_v/CNT电极对乙醇氧化的电催化性能。同时,

就 MoO_x 沉积量对电极的催化活性和乙醇氧化活化能的影响进行了研究。 $Pt/MoO_x/CNT$ 电极的长期循环稳定性研究表明,经过500圈循环伏安扫描后, $Pt/MoO_x/CNT$ 电极的催化活性仍保持在最初的82%,而Pt/CNT和Pt-Ru/CNT电极分别为55%和72%。

关键词 乙醇氧化,碳纳米管,氧化钼,铂

分类号

Preparation of Pt-MoO_r/CNT Electrode and Its Electrocatalytic Property for Ethanol Electrooxidation

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Abstract MoO_x , a non-noble secondary catalyst instead of noble metal Ru, was dispersed on carbon nanotubes (CNT) and then the primary catalyst Pt was electrodeposited on MoO_x /CNT/graphite electrode by potential-step method. CNT used was grown directly on graphite disk by chemical vapor deposition. The Pt-MoO $_x$ /CNT/graphite electrode was characterized by scanning electron microscopy and energy dispersive X-ray spectroscopy. The electrocatalytic property and the long-term cyclic stability of Pt-MoO $_x$ /CNT/graphite electrode for ethanol oxidation have been investigated in 1.0 mol/L $CH_3CH_2OH + 0.1$ mol/L H_2SO_4 by cyclic voltammetry. On the other hand, the effect of the loading mass of MoO_x on specific activity of Pt-MoO $_x$ /CNT/graphite electrode and the activation energy of ethanol oxidation were investigated. The Pt-MoO $_x$ /CNT/graphite electrode shows good long-term cyclic stability. 82% value of peak current density at the oxidation peak around 0.69 V (vs. SCE) remains after 500 cycles, which is higher than those of Pt/CNT/graphite and Pt-Ru/CNT/graphite electrodes with the corresponding values of 55% and 72%, respectively.

Key words ethanol oxidation carbon nanotube molybdenum oxide platinum

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扩展功能

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