杂多阴离子的XPS研究 II:Keggin结构钨钼杂多酸热稳定性差异的研究

王占文,吴乃钧,方景礼

南京化工学院应化系;南京大学大学系

收稿日期 修回日期 网络版发布日期 接受日期

摘要 加入优级纯CsCl作内标,用其中Cs3d5/2校正荷电效应,准确地测定了HPWnMo12-n(n=0,3,6,9,12)及HXM12 (X=P,As,Si,Ge;M=W,Mo)中Ols及HPM12中P2p的结合能,发现钨系中Ols的结合能比钼系中的高约 0.5eV;P2p的结合能则相应地高0.8eV;同种金属的磷系杂多酸中Ols的结合能比其他杂原子中的高 0.3eV。通过差热分析发现同类型杂多酸的热稳定性与阴离子中的Ols结合能呈近似平行的关系。结果表明, 跟钼系杂多酸相比,钨系杂多酸中氧原子有较低的电子云密度,钨氧之间有更多的轨道重叠; 含磷杂多酸跟其他杂多酸相比,金属、氧之间有更强的相互作用。

 关键词
 红外分光光度法
 杂多酸
 钼酸盐
 磷酸盐
 络合物
 内标
 热稳定性
 电子密度
 差热分析

 钨酸盐
 结合能
 结合能
 X射成光电子谱法

分类号 0657 064

# Studies of heteropolyanion by XPS.II.SPS study on the difference of thermostability between 12-heterotungstic and molybdic acids

WANG ZHANWEN, WU NAIJUN, FANG JINGLI

Abstract The binding energies of oxygen 1s and heteroatoms (P) in PWnMo12-n (n = 0, 3, 6, 9, 12) and XM12 (X = P, Asw, Si, Ge; M = W, Mo) were determine by adding highly pure CsCl as an internal label using Cs3d5/2 to calibrate the charge effect. The O1s and P binding energies in heterotungstic acids were higher by 0.5 eV and 0.8 eV, resp., than those in heteromolybdic acids, and the O1s binding energies in PM12 were 0.3 eV higher than those in other heteroatom's. The thermal stability of the heteropoly acids was closely paralleled to the variation of the O1s binding energy. Compared with XMo12, the electron d. of oxygen was lower and the overlap of the orbits between M and O was more effective in XW12, and that the interreaction between M and O in PM12 was stronger than that in other XM12.

Key wordsINFRARED SPECTROPHOTOMETRYHETEROPOLYACIDMOLYBDATEPHOSPHATECOMPLEX COMPOUNDSINTERNAL STANDARDTHERMAL STABILITYELECTRON DENSITYDIFFERENTIAL THERMAL ANALYSISTUNGSTATEBINDING ENERGYBINDING ENERGY

DOI:

通讯作者

#### 扩展功能

### 本文信息

- ► Supporting info
- ▶ **PDF**(0KB)
- ▶[HTML全文](0KB)
- ▶参考文献

## 服务与反馈

- ▶把本文推荐给朋友
- ▶加入我的书架
- ▶加入引用管理器
- ▶复制索引
- Email Alert
- ▶ 文章反馈
- ▶ 浏览反馈信息

# 相关信息

- ▶ <u>本刊中 包含"红外分光光度法"的</u> 相关文章
- ▶本文作者相关文章
- 王占文
- · 吴乃钧
  - 方景礼