Al2O3自电解质溶液中吸附CrO4^2^-的特性研究

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摘要 本文特别延长了pH范围(2-10),详细研究pH、温度、外加正负离子对CrO4^2^-吸附量的影响,说明这种吸附主要是以离子交换方式进行的,并认为表面活性点是产生这类吸附的重要原因.

] 氧化铝 吸附 离子交换 铬酸盐 P 电解质 表面活性 阴离子

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Characteristic study of adsorption of chromate anion from electrolytic solution on alumina

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Abstract The adsorption isotherms of chromate (CrO42-) from electrolytic solution (0.01M NaCl) on Al2O3 were determine at various pH (2-10) and concns. Max. adsorption occurs at pH u4.5. Adsorption of chromate decreases to 0 near the PZC (pH 9). The presence of some anions tends to decrease chromate adsorption. The order is citrate > PO43- > SO42- > NO3-. The presence of some cation tends to increase chromate adsorption. The order is Ba2+ > Ni2+ > K+. Chromate adsorption is not influenced by temperature These results are explained and the adsorption mechanism is discussed.

Key words ALUMINIUM OXIDE ADSORPTION ION EXCHANGING CHROMATE P ELECTROLYTE SURFACE ACTIVITY ANION

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