Full Paper

铁氰化铜修饰陶瓷碳复合电极的制备、电化学性质及其电催化性能

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摘要 采用两步溶胶-

凝胶技术制备了铁氰化铜修饰陶瓷碳复合电极。研究了该修饰电极的电化学性质和电催化活性。结果表明:该修饰电极在 $0.4\sim1.0$ V电位范围内产生一对准可逆的表面控制的氧化还原峰,电极反应电子传递速率常数(k.)为 $10.37s^{-1}$,式量电位 E^0 为 $0.682V(vs.\ SCE)$,且对 K^+ 离子活度具有能斯特响应。该修饰电极对肼的氧化具有很好的电催化作用,

异相催化反应速率常数为3.29×10³ M⁻¹·s⁻¹。安培法检测肼的线性范围为3.0×10⁻⁶~7.5×10⁻⁴mol·dm⁻³,检出限为8.0×10⁻⁷ mol·dm⁻ 3。该修饰电极具有修饰程序简单、表面更新容易、稳定性好、对肼的响应快和线性范围宽等特点,有望用于化学试剂、 工业产品和生物样品中肼含量的测定。

铁氰化铜,复合陶瓷碳电极,电催化氧化,肼,安培检测 关键词

分类号

Preparation, Electrochemical Behavior and Electrocatalytic Activity of a Copper Hexacyanoferrate Modified 本文作者相关文章 Ceramic Carbon Electrode

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Abstract A copper hexacyanoferrate modified ceramic carbon electrode (CuHCF/CCE) had been prepared by two-step sol-gel technique and characterized using electrochemical methods. The resulting modified electrode showed a pair of well-defined surface waves in the potential range of 0.40 to 1.0 V with the formal potential of 0.682 V (vs. SCE) in 0.050 mol·dm⁻³ HOAc-NaOAc buffer containing 0.30 mol \cdot dm⁻³ KCl. The charge transfer coefficient (α) and charge transfer rate constant (k_a) for the modified electrode were calculated. The electrocatalytic activity of this modified electrode to hydrazine was also investigated, and chronoamperometry was exploited to conveniently determine the diffusion coefficient (D) of hydrazine in solution and the catalytic rate constant (k_{cat}). Finally, hydrazine was determined with amperometry using the resulting modified electrode. The calibration plot for hydrazine determination was linear in 3.0×10^{-6} — 7.5×10^{-4} mol·dm⁻³ with the detection limit of 8.0×10^{-7} mol·dm⁻³. This modified electrode had some advantages over the modified film electrodes constructed by the conventional methods, such as renewable surface, good long-term stability, excellent catalytic activity and short response time to hydrazine.

Key words copper hexacyanoferrate ceramic carbon electrode chronoamperometry electrocatalytic oxidation hydrazine amperometry

DOI:

扩展功能

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