

Full Papers

1,1-环丁烷二羧酸单、双核铜配合物的合成、晶体结构、热分析和磁性李明星^a, 戴惠¹, 邵敏¹, 石磊¹, 林昆华¹, 程智萱¹, 翁林红²¹上海大学理学院化学系, 上海200444, 中国²复旦大学化学系, 上海200433, 中国

收稿日期 2005-8-17 修回日期 2005-12-27 网络版发布日期 接受日期

摘要 1,1-环丁烷二羧酸根 (cbdc) 与Cu(ClO₄)₂和邻啡罗啉 (phen) 在乙醇-水溶液中反应, 合成二个铜配合物: [Cu(cbdc)(phen)(H₂O)]·2H₂O (**1**) 和[Cu₂(cbdc)(phen)₂(H₂O)₂](ClO₄)₂·H₂O (**2**)。晶体**1** 属单斜晶系, *P*2(1)/*c*空间群, 晶胞参数 $a=0.9428(4)$, $b=1.2183(5)$, $c=1.6265(7)$ nm, $\beta=102.418(5)^\circ$, $V=1.8246(13)$ nm³, $Z=4$, $R=0.0445$, $wR_2=0.0947$ 。铜离子被cbdc、phen和H₂O五配位, 形成四方锥结构。分子间通过 $\pi\cdots\pi$ 重叠和氢键作用, 堆积成三维超分子结构。采用TG-DSC热分析详细研究了该配合物的热分解过程。晶体**2** 属单斜晶系, *P*2(1)/*c*空间群, 晶胞参数 $a=0.8897(3)$, $b=1.9130(8)$, $c=1.9936(8)$ nm, $\beta=99.04(2)^\circ$, $V=3.351(2)$ nm³, $Z=4$, $R=0.0540$, $wR_2=0.1102$ 。该化合物为羧基桥联的双核铜配合物, Cu(1)被cbdc和phen平面四配位, Cu(2)被cbdc、phen和二分子水五配位, 形成变形的三角双锥结构。变温磁化率研究(2-300 K)表明, 该配合物的磁性遵从居里定律, 对该配合物的磁性和结构关系做了讨论。

关键词 [铜配合物](#)、[1,1-环丁烷二羧酸](#)、[晶体结构](#)、[热稳定性](#)、[磁性](#)。

分类号

Synthesis, Crystal Structures, Thermal Analysis and Magnetic Property of Mono- and Binuclear 1,1-Cyclobutanedicarboxylate Copper ComplexesLI Ming-Xing^{a,1}, DAI Hui¹, SHAO Min¹, SHI Lei¹, LIN Kun-Hua¹, CHENG Zhi-Xuan¹, WENG Lin-Hong²¹ Department of Chemistry, College of Science, Shanghai University, Shanghai 200444, China² Department of Chemistry, Fudan University, Shanghai 200433, China

Abstract Two new copper complexes, [Cu(cbdc)(phen)(H₂O)]·2H₂O (**1**) and [Cu₂(cbdc)(phen)₂(H₂O)₂](ClO₄)₂·H₂O (**2**) (cbdc = 1,1-cyclobutanedicarboxylate and phen = 1,10-phenanthroline), were synthesized by reaction of cbdc with Cu(ClO₄)₂ and phen in ethanol aqueous solution. Complex **1** crystallizes in monoclinic system with space group *P*2(1)/*c* and $a=0.9428(4)$ nm, $b=1.2183(5)$ nm, $c=1.6265(7)$ nm, $\beta=102.418(5)^\circ$, $V=1.8246(13)$ nm³, $Z=4$, $R=0.0445$, $wR_2=0.0947$. The structure of **1** is discretely mononuclear, which is packed by $\pi\cdots\pi$ interaction forming a 3D supramolecular structure where Cu(II) ion is five-coordinated and has square-pyramidal coordination geometry. Its thermal decomposition procedure detail was studied by thermal analysis TG-DSC. Complex **2** belongs to monoclinic system with space group *P*2(1)/*c* and $a=0.8897(3)$ nm, $b=1.9130(8)$ nm, $c=1.9936(8)$ nm, $\beta=99.04(2)^\circ$, $V=3.351(2)$ nm³, $Z=4$, $R=0.0540$, $wR_2=0.1102$. The structure of **2** is a discrete binucleus, where Cu(1) is four-coordinated by phen and cbdc in a square-planar geometry while Cu(2) is five-coordinated by phen, one O of cbdc and two H₂O, which can be best described as distorted trigonal-bipyramidal geometry. Cu(1) and Cu(2) are linked by carboxylic group of cbdc in a bidentate bridging fashion. Variable-temperature magnetic susceptibilities of **2** in 2—300 K showed that its magnetic behavior obeyed Curie law.

Key words [copper complex](#) [1,1-cyclobutanedicarboxylate](#) [crystal structure](#) [thermal analysis](#) [magnetic property](#)

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