

[Ho₂(phen)₄(H₂O)₄(OH)₂](phen)₂(NO₃)₄的合成、晶体结构及磁性

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摘要 在甲醇与水的混合溶剂中,经浓硝酸硝化的Ho₂O₃与1,10-邻菲罗啉反应形成氢氧根桥联的双核钬配合物[Ho₂(phen)₄(H₂O)₄(OH)₂](phen)₂(NO₃)₄ (phen = 1,10-邻菲罗啉)。单晶X射线衍射晶体结构测定表明晶体属三斜晶系, P1-bar (no. 2)空间群,晶胞参数a = 1.1241(1) nm, b = 1.1439(1) nm, c = 1.4058(1) nm, α = 93.989(7)°, β = 98.173(7)°, γ = 108.19(1)°, V = 1.6874(4) nm³, Z = 1, D_c = 1.737 g/cm³, F(000) = 880, 7752个独立衍射点中, 5702个可观测点满足F_o-2 ≥ 2σ(F_o-2), R₁ = 0.0499, wR₂ = 0.858。标题配合物由中心对称的双核[Ho₂(phen)₄(H₂O)₄(OH)₂](4+)配阳离子, 邻菲罗啉phen分子及硝酸根NO₃⁻阴离子组成。每个稀土原子与2个邻菲罗啉配体, 2相水分子和2个氢氧根配位形成配位数为8的[HoN₄O₄]四方反棱柱。配位多面体通过两氢氧根基团形成共棱的[Ho₂N₈O₆]双四方反棱柱[d(Ho-N) = 0.2549~0.2565 nm, d(Ho-O(H₂O)) = 0.2356, 0.2366 nm, d(Ho-O(OH)) = 0.2223, 0.2240 nm]。通过氢键和芳环堆积作用, 配阳离子和邻菲罗啉分子排列形成平行于(10 1-bar)的两维层结构, NO₃⁻阴离子位于层之间。标题配合物为顺磁物质, 在5~300K区间内遵循Curie-Weiss定律X_m(T + 4.43) = 14.72 cm³·K·mol⁻¹(X_m为每摩尔Ho~(3+)离子磁化率), 其Ho~(3+)离子的室温有效磁矩为 10.76 B. M., 与Ho~(3+)自由离子的磁矩基本相同, 表明稀土离子间不存在磁交换作用。

关键词 [钬络合物](#) [双核络合物](#) [晶体结构](#) [磁性](#) [二氮杂菲 P](#)

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Synthesis, Crystal Structure and Magnetic Property of [Ho₂(phen)₄(H₂O)₄(OH)₂](phen)₂(NO₃)₄

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Abstract Reaction of nitrated Ho₂O₃ with 1,10-phenanthroline in CH₃OH/H₂O results in a novel hydroxo bridged dinuclear Ho(III) complex [Ho₂(phen)₄(H₂O)₄(OH)₂](phen)₂(NO₃)₄. Single crystal X-ray diffraction study shows that the title compound crystallizes in the triclinic space group P 1-bar (no. 2) with the cell dimensions; a = 1.1241(1) nm, b = 1.1439(1) nm, c = 1.4058(1) nm, α = 93.989(7)°, β = 98.173(7)°, γ = 108.19(1)°, V = 1.6874(4) nm³, Z = 1, D_c = 1.737 g/cm³, F(000) = 880, R₁ = 0.0499, wR₂ = 0.0858 for 5702 observed reflections [F_o-2 ≥ 2σ(F_o-2)] out of 7752 unique reflections. The compound consists of the centrosymmetric dinuclear [Ho₂(phen)₄(H₂O)₄(OH)₂](4+ complex cations, uncoordinated phen molecules and nitrate anions. The holmium atom coordinates to two phen ligands, two H₂O molecules and two hydroxo groups to form a [HoN₄O₄] square antiprism. An edge-shared bi-square-antiprism forms via the two bridging OH groups [d(Ho-N) = 0.2549~0.2565 nm, d(Ho-O(H₂O)) = 0.2356, 0.2366 nm, d(Ho-O(OH)) = 0.2223, 0.2240 nm]. The complex cations and the uncoordinating phen molecules are assembled via hydrogen bonding and π-π stacking interactions into 2D layers parallel to (10 1-bar), between which the nitrate anions are sandwiched. The title compound behaves paramagnetically and obeys Curie-Weiss law X_m(T + 4.43) = 14.72 cm³·K·mol⁻¹ between 5~300 K (X_m: magnetic susceptibility per mole Ho~(3+) ions). The effective magnetic moment at room temperature is 10.76 B. M., which is equal to the expected value for a free Ho~(3+) ion, suggesting no magnetic exchange within the dinuclear dinuclear cations.

Key words [HOLMIUM COMPLEX](#) [DINUCLEAR COMPLEX](#) [CRYSTAL STRUCTURE](#) [MAGNETISM](#) [PHENANTHROLINE P](#)

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