

[本期目录](#) | [下期目录](#) | [过刊浏览](#) | [高级检索](#)[\[打印本页\]](#) [\[关闭\]](#)**论文****官能团化己内酯与丙交酯无规共聚物的合成与降解性能**戴炜枫¹, 杜征臻¹, 何月英¹, 郎美东^{1,2}1. 华东理工大学材料科学与工程学院, 超细材料制备与应用教育部重点实验室, 上海 200237;
2. 复旦大学聚合物分子工程教育部重点实验室, 上海 200433**摘要:**

研究了官能团化新型己内酯单体的合成及其与丙交酯无规共聚物的降解性能。首先, 环己酮和N-异丙基丙烯酰胺通过Michael加成反应合成了2-(N-异丙基酰胺乙烯基)-环己酮; 然后, 以间氯过氧化苯甲酸为氧化剂, 通过Baeyer-Villiger氧化反应, 制备带有酰胺官能团的己内酯单体6-(N-异丙基酰胺乙烯基)-ε-己内酯; 最后, 在异辛酸亚锡[Sn(Oct)₂]的催化下与丙交酯开环聚合, 得到新型己内酯与丙交酯的无规共聚物。采用¹H NMR, SEC和DSC表征了聚合物的结构和热力学性能。同时通过黏度法、失重法和SEM对该聚合物的降解性能进行了表征。结果表明, 该共聚物的降解速率明显增快, 材料降解2个月后, 材料的质量损失达到28.1%, 特性黏度降低近40%。

关键词: 聚己内酯; 聚丙交酯; 侧基官能团; 生物可降解材料; 无规共聚物**Synthesis, Characterization and Degradability of P(LA-co-ACL) Copolymer Bearing Pendant N-Substituted Acylamine Group**DAI Wei-Feng¹, DU Zheng-Zhen¹, HE Yue-Ying¹, LANG Mei-Dong^{1,2*}1. Key Laboratory for Ultrafine Materials, Ministry of Education, School of Materials Science and Engineering, East China University of Science and Technology, Shanghai 200237, China;
2. Key Laboratory of Molecular Engineering of Polymers, Ministry of Education, Fudan University, Shanghai 200433, China**Abstract:**

The synthesis, characterization and polymerization of a new cyclic 6-(N-isopropyl acylamine ethyldene)-ε-caprolactone(ACL) were reported. α-Substituted cyclohexanone was synthesized by Michael reaction of N-isopropylacrylamide and cyclohexanone. Then it was subsequently converted into ACL by the Baeyer-Villi-ger oxidation reaction using of 3-chloroperoxybenzoic acid(*m*-CPBA) as the oxidant. This new lactone was copolymerized with various contents of *L*-LA by bulk ring-opening polymerization(ROP) initiated by Sn(Oct)₂. All the copolymers were characterized by ¹H NMR, SEC and DSC. The copolymer formed flexible films and was used to study its degradability. A phosphate buffer (pH=7.4) with temperature 37 °C was adopted to proceed the degrading study all through. The hydrolytic degradation of P(LA-co-ACL) was much faster, which is confirmed by the mass loss and change of intrinsic viscosity.

Keywords: Poly(ε-caprolactone); Poly(*L*-lactide); Pendant group; Biodegradable material; Random copolymer

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