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Production

Subscription

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■ Volumes and Issues
■ Contents of Issue 9

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Kinetic isotope effects in the gas phase reactions of OH and CI with CH_3CI , CD_3CI , and $^{13}CH_3CI$

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Abstract. The kinetic isotope effects in the reactions of CH₃CI, ¹³CH₃CI and CD₃CI with OH radicals and CI atoms were studied in relative rate experiments at 298±2 K and 1013±10 mbar. The reactions were carried out in a smog chamber using long path FTIR detection and the spectroscopic data analyzed employing a non-linear least squares spectral fitting method using measured high-resolution infrared spectra as well as absorption cross sections from the HITRAN database. The reaction rates of $^{13}\mathrm{CH_2CI}$ and $\mathrm{CD_2CI}$ with OH and CI were determined relative to $\mathrm{CH_3CI}$ as: $k_{\text{OH+CH3CI}}k_{\text{OH+CH3CI}}/k_{\text{OH+13CH3CI}} k_{\text{OH+13CH3CI}} = 1.059 \pm 0.008,$ $k_{\text{OH+CH3CI}}k_{\text{OH+CH3CI}}/k_{\text{OH+CD3CI}}k_{\text{OH+CD3CI}} = 3.9 \pm 0.4$ $k_{\text{CI+CH3CI}} k_{\text{CI+CH3CI}} / k_{\text{CI+13CH3CI}} k_{\text{CI+13CH3CI}} = 1.070 \pm 0.010$ and $k_{\rm CI+CH3CI}k_{\rm CI+CH3CI}/k_{\rm CI+CD3CI}k_{\rm CI+CD3CI}=4.91\pm0.07.$ The uncertainties given are 2σ from the statistical analyses and do not include possible systematic errors. The unexpectedly large ¹³C kinetic isotope effect in the OH reaction of CH₃CI has important implications for the global emission inventory of CH₂CI.

■ <u>Final Revised Paper</u> (PDF, 1392 KB) ■ <u>Supplement</u> (39 KB) <u>Discussion</u> <u>Paper</u> (ACPD)

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