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# 298 K rate coefficients for the reaction of OH with i - $C_3H_7I$ , $n - C_3H_7I$ and $C_3H_8$

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Abstract. The kinetics of the title reactions were investigated using the laser photolysis - resonance fluorescence method, employing the sequential two-photon dissociation of  $NO_2$  in the presence of  $H_2$  as the OH source. The 298 K rate constant for OH +  $C_3H_8$  was found to be (1.15  $\pm$  $0.1) \times 10^{-12}$  cm<sup>3</sup> s<sup>-1</sup>, in excellent agreement with the literature recommendation, and with a separate determination using HNO3 photolysis at 248 nm as the OH source. The 298 K rate constants for OH +  $n - C_3H_7I$  and  $i - C_3H_7I$  were measured for the first time and found to be  $(1.47 \pm 0.08)$  and  $(1.22 \pm 0.06) \times 10^{-12}$  cm<sup>3</sup> s<sup>-1</sup>, respectively. The errors include an assessment of systematic error due to concentration measurement, which, for the propyl-iodides was minimised by on-line UVabsorption spectroscopy. These results show that reaction with OH is an important sink for  $n - C_3H_7I$  and  $i - C_3H_7I$ , which has implications for the reactive iodine budget of the marine boundary layer.

■ Final Revised Paper (PDF, 116 KB)
■ Discussion Paper (ACPD)

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