## Structural Fluorine in Sepiolite

## Julio Santaren, Jesus Sanz and Eduardo Ruiz-Hitzky

Research and Development Department, Tolsa, S.A. P.O. Box 38017, 28080 Madrid, Spain Instituto de Ciencia de Materiales, C.S.I.C. Serrano 115 bis, 28006 Madrid, Spain

**Abstract:** Sepiolite from Vallecas-Vicálvaro, Spain, contains 1.3% fluorine. Laser microprobe mass spectrometry of this sepiolite suggests the presence of fragments of  $(SiO_2)_nOMgF$  and  $(SiO_2)_nOMgOH$ , which are typical of the sepiolite structure. During thermal dehydroxylation, the fluorine in this sepiolite is removed simultaneously with OH groups at about 750° C. Nuclear magnetic resonance spectroscopy (NMR) of <sup>19</sup>F indicates that the fluorine is located in the interior of the sepiolite structure, probably substituting for OH groups, and is homogeneously distributed. In the Vallecas-Vicálvaro sepiolite, about one of every four OH groups bound to  $Mg^{2+}$  is substituted by fluorine. The kinetics of extraction of  $Mg^{2+}$  and  $F^-$  ions by acid treatment (1 N HCl) shows a more rapid extraction of  $Mg^{2+}$ , with a monotonous decrease of the Mg/F ratio as the extent of extraction increases. These results support the internal location of the fluorine, as suggested by the NMR data.

**Key Words:** Fluorine • Hydroxyl • Laser microprobe mass spectrometry • Dissolution • Nuclear magnetic resonance • Sepiolite

Clays and Clay Minerals; February 1990 v. 38; no. 1; p. 63-68; DOI: <a href="https://doi.org/10.1346/CCMN.1990.0380109">10.1346/CCMN.1990.0380109</a> © 1990, The Clay Minerals Society (<a href="https://www.clays.org">www.clays.org</a>)