

---

# Evolution of Hematite Surface Microtopography Upon Dissolution by Simple Organic Acids

Patricia A. Maurice<sup>1</sup>, Michael F. Hochella Jr.<sup>1</sup>, George A. Parks<sup>1</sup>, Garrison Sposito<sup>2</sup> and Udo Schwertmann<sup>3</sup>

<sup>1</sup> Department of Geological and Environmental Sciences Stanford University, Stanford, California 94305

<sup>2</sup> Department of Environmental Science, Policy, and Management University of California at Berkeley, Berkeley, California 94720

<sup>3</sup> Institute of Soil Science, Technical University of Munich, Germany

**Abstract:** The surface microtopography of hematite over the course of dissolution in oxalic and citric acids was examined by *in-situ* and *ex-situ* atomic-force microscopy, *In-situ* imaging of the basal-plane surface of a centimeter-scale natural hematite sample immersed in 2 mM citric acid demonstrated that the basal-plane surface was relatively unreactive; rather, dissolution occurred along step edges and via etch-pit formation. *Ex-situ* imaging of synthetic hematite particles following batch dissolution in 1 mM oxalic acid showed similar dissolution features on basal-plane surfaces; in addition, etching along particle edges was apparent. The presence of etch features is consistent with a surface-controlled dissolution reaction. The results are in agreement with previous investigations suggesting that the basal-plane surface is relatively unreactive with respect to ligand exchange. Both *in-situ* and *ex-situ* imaging of particle surfaces can provide valuable information on the roles of surface structures and microtopographic features in mineral dissolution.

**Key Words:** Atomic force microscope • Clay mineral surfaces • Dissolution • Hematite • Organic acids

*Clays and Clay Minerals*; February 1995 v. 43; no. 1; p. 29-38; DOI: [10.1346/CCMN.1995.0430104](https://doi.org/10.1346/CCMN.1995.0430104)

© 1995, The Clay Minerals Society

Clay Minerals Society ([www.clays.org](http://www.clays.org))

---