Formation of Spinel from a Hydrotalcite-Like Compound at Low Temperature: Reaction between Edges of Crystallites

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Abstract: The thermal decomposition behavior of hydrotalcite-like compounds (HTlcs) prepared by reconstruction of calcined HTlcs is described. From the results of X-ray diffraction (XRD), it seems that dicarboxylate intercalates of HTlc calcined at 500 $^{\circ}$ C are completely reconstructed to Mg-Al-CO₃ HTlc by exposure to aqueous Na₂CO₃. However, the Mg-Al-CO₃ HTlc reconstructed under particular conditions yields spinel (MgAl₂O₄) at 400 $^{\circ}$ C. This temperature is very low, because Mg-Al-CO₃ HTlc that has been reported yields spinel at 900 $^{\circ}$ C after forming a Mg-Al double oxide. The reconstructed Mg-Al-CO₃ HTlc that yields spinel at 400 $^{\circ}$ C is obtained when the following conditions are fulfilled: the crystallites of the starting dicarboxylate intercalates are coagulated tightly and the calcined HTlcs and reconstructed materials are not ground. The Mg-Al-CO₃ HTlc reconstructed under these conditions contains only 55– 70% of carbonate anions required by stoichiometry. Therefore, we conclude that the transformation of reconstructed Mg-Al-CO₃ HTlc to spinel at 400 $^{\circ}$ C is the result of a reaction occurring between edges of crystallites.

Key Words: Double Hydroxide • Grinding • Hydrotalcite • Reconstruction • Spinel • Thermal Decomposition

Clays and Clay Minerals; December 1997 v. 45; no. 6; p. 842-853; DOI: <u>10.1346/CCMN.1997.0450608</u> © 1997, The Clay Minerals Society Clay Minerals Society (<u>www.clays.org</u>)