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# The Al Pillaring of Clays. Part II. Pillaring with $[\text{Al}_{13}\text{O}_4(\text{OH})_{24}(\text{H}_2\text{O})_{12}]^{7+}$

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**Abstract:** Hectorite and saponite are exchanged with  $[\text{Al}_{13}\text{O}_4(\text{OH})_{24}(\text{H}_2\text{O})_{12}]^{7+}$  and the amount of  $\text{Al}^{3+}$  adsorbed and  $\text{Na}^+$  released are followed as a function of the exchange conditions. On saponite the reaction is a pure ion exchange with 2– 2.15 mmol  $\text{Al}^{3+}/\text{g}$  adsorbed and release of 0.80 mmol  $\text{Na}^+/\text{g}$ . On hectorite the ion exchange is accompanied by supplementary hydrolysis-polymerization of  $\text{Al}_{13}$ . When excess Al is offered in the form of  $\text{Al}_{13}$ , ion exchange is incomplete and is accompanied by precipitation and polymerization of  $\text{Al}_{13}$  on the surface of both hectorite and saponite. The typical spacing of 1.8 nm is developed after washing, when at least 1.3– 1.4 mmol  $\text{Al}^{3+}/\text{g}$  is adsorbed. Above a loading of 2.2– 2.5 mmol/g the 1.8 nm spacing is obtained without washing. Only pillared saponite with a loading of at least 1.9 mmol  $\text{Al}^{3+}/\text{g}$  is thermally stable up to 550° C.

**Key Words:** Aluminium • Hectorite • Keggin ion • Pillaring • Saponite

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