

---

# Nuclear Magnetic Resonance, Infrared, and X-ray Powder Diffraction Study of Dimethylsulfoxide and Dimethylselenoxide Intercalates with Kaolinite

M. Raupach, P. F. Barron and J. G. Thompson

CSIRO—Division of Soils, Private Bag No. 2, Glen Osmond South Australia 5064, Australia  
Brisbane NMR Centre, School of Science, Griffith University Nathan, Queensland 4111, Australia  
Research School of Chemistry, Australian National University GPO Box 4, Canberra, A.C.T. 2605, Australia

**Abstract:** Dimethylselenoxide (DMSeO) forms three structurally resolvable intercalates with kaolinite ( $d(001) = 10.95, 11.26,$  and  $11.38 \text{ \AA}$ ). The  $11.26\text{-\AA}$  kaolinite: DMSeO intercalate is structurally analogous to the 3-D ordered kaolinite: DMSO intercalate ( $d(001) = 11.22 \text{ \AA}$ ). Infrared and solid-state  $^{77}\text{Se}$  nuclear magnetic resonance data indicate that all DMSeO molecules are equivalent in the structure and, therefore, that the  $11.26\text{-\AA}$  kaolinite: DMSeO intercalate structure is C-face centered. Structural model refinement from X-ray powder diffraction (XRD) data further support this conclusion ( $P1, a = 5.195(2), b = 8.990(4), c = 11.946(5) \text{ \AA}, \alpha = 91.33(2)^\circ, \beta = 109.39(2)^\circ, \gamma = 89.77(2)^\circ$ ). The kaolinite: DMSO intercalate structure was subsequently re-refined from the XRD profile in C-face centered  $P1$ . The derived orientation of the DMSO and DMSeO molecules with respect to the basal plane of their respective intercalates is in agreement with polarized infrared measurements of the angles with  $ab$  of  $\text{S=O}$  ( $40.3^\circ$ ) and  $\text{Se=O}$  ( $38.8^\circ$ ) for the  $11.26\text{-\AA}$  intercalate. The locations of the organic molecules also agree with observed infrared band splittings and perturbations. Interatomic distances calculated from the band shifts agree with those for the XRD derived structures.

The  $11.38\text{-\AA}$  kaolinite: DMSeO intercalate is closely related to the  $11.26\text{-\AA}$  intercalate, the main differences being a 2-fold disorder in the orientation of the DMSeO molecule and less penetration of the kaolinite ditrigonal cavity by that molecule. The  $10.95\text{-\AA}$  kaolinite: DMSeO intercalate, displaying disorder parallel to  $[110]$ , was obtained from the  $11.26\text{-}$  or  $11.38\text{-\AA}$  intercalates by removal of some DMSeO.

**Key Words:** Dimethylselenoxide • Dimethylsulfoxide • Infrared spectroscopy • Intercalate • Kaolinite • Nuclear magnetic resonance • X-ray powder diffraction

*Clays and Clay Minerals*; June 1987 v. 35; no. 3; p. 208-219; DOI: [10.1346/CCMN.1987.0350307](https://doi.org/10.1346/CCMN.1987.0350307)  
© 1987, The Clay Minerals Society  
Clay Minerals Society ([www.clays.org](http://www.clays.org))

---