



Potential role of copper availability in nitrous oxide accumulation in a temperate lake

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ABSTRACT: Denitrifying bacteria require copper for the synthesis of nitrous oxide reductase. In the absence of sufficient bioavailable Cu, nitrous oxide (N₂O) may accumulate in natural waters during denitrification. Cultures of *Paracoccus denitrificans* and natural bacterial assemblages collected from a mesotrophic lake (Linsley Pond) were grown at varying Cu concentrations to determine the Cu speciation that results in elevated N₂O accumulation. *P. denitrificans* experienced Cu limitation beginning at inorganic Cu concentrations of 0.6 fmol L⁻¹ (-log of inorganic Cu, pCuⁱ, = 15.2). The natural community did not show an effect until pCuⁱ was reduced to 23.7 with 8-hydroxyquinoline, resulting in an approximate 10-fold increase in N₂O concentrations during denitrification. Additions of ethylenediaminetetraacetic acid alone to natural communities did not affect N₂O concentrations. Natural copper-binding ligands detected with competitive ligand exchange-cathodic stripping voltammetry occurred at concentrations of 6 to 25 nmol L⁻¹ with conditional stability constants (K_{CuL}^{\prime}) between 10^{14.4} and 10^{15.1}. Although more than 99% of total Cu in Linsley Pond was bound to these ligands, inorganic Cu concentrations remained 10 orders of magnitude above those found to increase N₂O accumulation during denitrification incubations. Measurements of nitrogen species, dissolved oxygen, and sulfide in the water column of Linsley Pond over the spring growing season revealed that N₂O was produced by assimilatory nitrate reduction and nitrification in addition to denitrification, with nitrification generating most of the N₂O found in the surface waters of the lake. The results suggest that inorganic Cu concentrations in Linsley Pond are sufficient to support denitrification. Moreover, some denitrifying bacteria may be able to access organically bound Cu, reducing the potential for this metal to affect N₂O production in other aquatic environments.

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