

论文摘要

中国有色金属学报

ZHONGGUO YOUSEJINSHUXUEBAO XUEBAO

第19卷 第10期 (总第127期) 2009年10月

 [PDF全文下载]  [全文在线阅读]

文章编号: 1004-0609(2009)10-1759-07

铝晶体自由表面的稳定性计算

张新明, 刘建才, 唐建国, 陈明安

(中南大学 材料科学与工程学院 有色金属材料科学与工程教育部重点实验室, 长沙 410083)

摘要: 采用第一原理赝势平面波方法, 计算铝晶体自由表面(100)、(110)和(111)的能量、几何与电子结构。根据表面能计算预测铝自由表面结构的稳定性。结果表明: 铝自由表面结构稳定性由强到弱的顺序为(111)、(100)、(110); 表面原子弛豫不仅引起表面几何结构的变化, 而且使表面层的电子结构与键合特性发生改变; (100)、(110)和(111)表层弛豫分别为3.337%、-6.147%和-2.364%; 表面电荷密度不同引起表面能差异, 表面原子层和次表面原子层的电荷面密度在s和p轨道上重新分布; 表面原子层电荷密度越大, 表面能越低。

关键字: Al 晶体; 第一原理; 表面弛豫; 表面能

Calculation of stability of free surfaces in aluminum crystal

ZHANG Xin-ming, LIU Jian-cai, TANG Jian-guo, CHEN Ming-an

(Key Laboratory of Non-ferrous Metal Materials Science and Engineering, Ministry of Education, School of Materials Science and Engineering, Central South University, Changsha 410083, China)

Abstract: The surface energy, atomic geometry and electronic structures of Al(100), (110) and (111) free surfaces were calculated using the method of supercell and the first-principles pseudopotential plane waves within generalized gradient approximation. According to the calculated surface energy, the structural stability of Al free surfaces from strong to weak is predicted in the order as (111), (100) and (110). The relaxation of the surface atom layers not only causes the change of geometrical structures of the surface models, but also leads to the variation of their electronic structures and bonding characters. For the (100), (110) and (111) free surfaces, the calculated surfaces relaxation are 3.337%, -6.147% and -2.364%, respectively. The surface energy is related to the surface electron density distribution, the electron density of orbital s and p of the first two surface atom layers redistributes. The higher the surface electron density is, the lower the surface energy is.

Key words: aluminum crystal; first principle; surface relaxation; surface energy

版权所有：《中国有色金属学报》编辑部 湘ICP备09001153号

地 址：湖南省长沙市岳麓山中南大学内 邮编： 410083

电 话： 0731-88876765, 88877197, 88830410 传真： 0731-88877197

电子邮箱： f-yssxb@mail.csu.edu.cn