

论文

熔融Sn—S/ZnCl<sub>2</sub>-NaCl体系界面脱硫反应动力学的阻抗分析

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摘要: 本文用交流阻抗方法研究了熔融Sn—S / ZnCl<sub>2</sub>-NaCl体系界面脱硫反应动力学过程,得到了该体系的阻抗响应图.通过用Randles等效电路,将所得的阻抗响应用非线性最小二乘法拟和,得到了Sn—S / ZnCl<sub>2</sub>-NaCl体系界面脱硫反应动力学的参数,溶液电阻Re,界面反应电荷传递电阻Rct,双电层电容Cd及界面脱硫反应的速率常数Kf.

关键词: 交流阻抗法 Randles等效电路 电荷传递电阻 双电层电容 渣金界面反应

THE IMPEDANCE ANALYSIS OF REACTION KINETICS AT THE INTERFACE OF MOLTEN Sn-S/ZnCl<sub>2</sub>-NaCl SYSTEM

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Abstract: The interfacial desulphurization reaction kinetics between molten Sn-S alloy and ZnCl<sub>2</sub>-NaCl salt was studied by an AC impedance method and the impedance response of this system was given. By using Randles equivalent circuit, the kinetic factors on electrode reaction between the alloy and the salt were obtained, such as electrolyte resistance Re, double-layer capacitance Cd, charge transfer resistance Rct, rate constant of desulfurization reaction Kf and so on. When overpotential η equals to -85 mV, these values are Re=2.93×10<sup>-5</sup> Ω.m<sup>2</sup>, Rct =2.47×10<sup>-5</sup> Ω.m<sup>2</sup>, Cd=1.01 F/m<sup>2</sup>, Kf=2.1 × 10<sup>-3</sup> s<sup>-1</sup>, respectively.

Keywords: AC impedance method Randles equivalent circuit charge-transfer resistance double-layer capacitance salt-metal interfacial reaction

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