

## Theoretical investigation of methane adsorption onto boron nitride and carbon nanotubes

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**Journal** Science and Technology of Advanced Materials [Create an alert](#) [RSS this journal](#)

**Issue** Volume 11, Number 4

**Citation** Masoud Darvish Ganji *et al* 2010 *Sci. Technol. Adv. Mater.* **11** 045001  
doi: [10.1088/1468-6996/11/4/045001](https://doi.org/10.1088/1468-6996/11/4/045001)

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**Abstract** Methane adsorption onto single-wall boron nitride nanotubes (BNNTs) and carbon nanotubes (CNTs) was studied using the density functional theory within the generalized gradient approximation. The structural optimization of several bonding configurations for a CH<sub>4</sub> molecule approaching the outer surface of the (8,0) BNNT and (8,0) CNT shows that the CH<sub>4</sub> molecule is preferentially adsorbed onto the CNT with a binding energy of -2.84 kcal mol<sup>-1</sup>. A comparative study of nanotubes with different diameters (curvatures) reveals that the methane adsorptive capability for the exterior surface increases for wider CNTs and decreases for wider BNNTs. The introduction of defects in the BNNT significantly enhances methane adsorption. We also examined the possibility of binding a bilayer or a single layer of methane molecules and found that methane molecules preferentially adsorb as a single layer onto either BNNTs or CNTs. However, bilayer adsorption is feasible for CNTs and defective BNNTs and requires binding energies of -3.00 and -1.44 kcal mol<sup>-1</sup> per adsorbed CH<sub>4</sub> molecule, respectively. Our *first-principles* findings indicate that BNNTs might be an unsuitable material for natural gas storage.

**PACS** [68.43.Mn Adsorption kinetics](#)  
[71.15.Mb Density functional theory, local density approximation, gradient and other corrections](#)  
[61.46.Fg Nanotubes](#)  
[71.15.Nc Total energy and cohesive energy calculations](#)

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**Dates** Issue 4 (August 2010)  
Received 24 十一月 2009, accepted for publication 21 七月 2010  
Published 10 九月 2010

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