

化学链制氢中 $\text{Fe}_2\text{O}_3/\text{LaFeO}_3$ 载氧体的性能研究

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Performance of $\text{Fe}_2\text{O}_3/\text{LaFeO}_3$ as oxygen carrier in chemical-looping hydrogen generation

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摘要 采用柠檬酸络合法制备 $\text{Fe}_2\text{O}_3/\text{LaFeO}_3$ 复合氧化物, 将该氧化物作为化学链制氢过程的载氧体, 在反应温度为 $900\text{ }^\circ\text{C}$ 、常压下, 对 $\text{Fe}_2\text{O}_3/\text{CH}_4$ (剂烷比)、进水量、金属负载量进行了考察。结果表明, 剂烷比为2: 1、进水量为0.1 mL、质量分数15%Fe时载氧体性能最好, 甲烷转化率达到60%, 单次循环氢气产量为45 mL。将该评价结果与XRD和 H_2 -TPR表征结果进行关联发现, 反应过程的活性位不是金属氧化物, 而是吸附氧, 而且吸附氧越容易还原, 甲烷转化率和氢气产量越高。通过连续60次还原-氧化循环发现, 该载氧体上甲烷转化率和氢气产量比较稳定, 循环后仍然保持钙钛矿结构。

关键词: 化学链制氢 载氧体 柠檬酸络合法 钙钛矿 甲烷

Abstract: $\text{Fe}_2\text{O}_3/\text{LaFeO}_3$ was prepared by citric acid complexation method and used as oxygen carrier in chemical-looping hydrogen generation; the effect of $\text{Fe}_2\text{O}_3/\text{CH}_4$ ratio, stream amount and metal loading on the performance of $\text{Fe}_2\text{O}_3/\text{LaFeO}_3$ as oxygen carrier at $900\text{ }^\circ\text{C}$ and atmospheric pressure were investigated. The results showed that the oxygen carrier performs best with CH_4 conversion of 60% and single-loop H_2 output of 45 mL, when the ratio of $\text{Fe}_2\text{O}_3/\text{CH}_4$, stream input and Fe loading are 2, 0.1 mL, 15%, respectively. The XRD and H_2 -TPR results suggest that the active site is the adsorbed oxygen rather than metal oxide; the easier for the reduction of the adsorbed oxygen, the higher are the conversion of CH_4 and the output of H_2 . Moreover, the oxygen carrier is still in perovskite structure and CH_4 conversion and H_2 output remain stable even after 60 cycles.

Key words: chemical-looping hydrogen generation oxygen carrier citric acid complexation method perovskite methane

收稿日期: 2013-03-07;

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引用本文:

梁皓, 宋喜军, 尹泽群等. 化学链制氢中 $\text{Fe}_2\text{O}_3/\text{LaFeO}_3$ 载氧体的性能研究[J]. 燃料化学学报, 2013, 41(12): 1512-1519.

LIANG Hao, SONG Xi-jun, YIN Ze-qun et al. Performance of $\text{Fe}_2\text{O}_3/\text{LaFeO}_3$ as oxygen carrier in chemical-looping hydrogen generation[J]. J Fuel Chem Technol, 2013, 41(12): 1512-1519.

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


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