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## Interaction of nitrogen dioxide with free base phthalocyanine

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Keywords

phthalocyanine, gas sensors, *ab initio* calculations

Abstract

Quantum chemical *ab initio* method was used to investigate interactions between  $\text{NO}_2$  and  $\text{H}_2\text{Pc}$ . Four possible sites for  $\text{NO}_2$  adsorption were considered. Full geometry optimisation together with frequency calculation was performed for the isolated molecules and the  $\text{NO}_2\text{-H}_2\text{Pc}$  complexes with the use of  $6\text{-}31g(d, p)$  and minimal basis sets, respectively. Two stable and two transition states were found. The  $\text{H}_2\text{Pc}$  molecular structure was found to be largely affected under the influence of  $\text{NO}_2$ . It was found that for a certain chemisorption site  $\text{H}_2\text{Pc}$  molecule might disintegrate and its building blocks could react with  $\text{NO}_2$  forming other species.



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