

# Enhanced Sonocatalytic Degradation of Rhodamine B by Graphene-TiO<sub>2</sub> Composites Synthesized by an Ultrasonic-Assisted Method

ZHU Lei, Trisha GHOSH, Chong-Yeon PARK, MENG Ze-Da, OH Won-Chun\*

Department of Advanced Materials Science &amp; Engineering, Hanseo University, Chungnam 356-706, Korea

ZHU Lei, Trisha GHOSH, Chong-Yeon PARK, MENG Ze-Da, OH Won-Chun\*

Department of Advanced Materials Science &amp; Engineering, Hanseo University, Chungnam 356-706, Korea

- 摘要
- 参考文献
- 相关文章

Download: PDF (637KB) [HTML](#) (1KB) Export: BibTeX or EndNote (RIS) Supporting Info

**摘要** A series of graphene-TiO<sub>2</sub> composites was fabricated from graphene oxide and titanium n-butoxide (TNB) by an ultrasonic-assisted method. The structure and composition of the nanocomposites were characterized by Raman spectroscopy, BET surface area measurements, X-ray diffraction, transmission electron microscopy, and ultraviolet-visible absorption spectroscopy. The average size of the TiO<sub>2</sub> nanoparticles on the graphene nanosheets was controlled at around 10–15 nm without using surfactant, which is attributed to the pyrolysis and condensation of dissolved TNB into TiO<sub>2</sub> by ultrasonic irradiation. The catalytic activity of the composites under ultrasonic irradiation was determined using a rhodamine B (RhB) solution. The graphene-TiO<sub>2</sub> composites possessed a high specific surface area, which increased the decolorization rate for RhB solution. This is because the graphene and TiO<sub>2</sub> nanoparticles in the composites interact strongly, which enhances the photoelectric conversion of TiO<sub>2</sub> by reducing the recombination of photogenerated electron-hole pairs.

**关键词:** [grapheme](#) [ultrasonication](#) [sonocatalytic degradation](#) [adsorption](#) [rhodamine B](#)

**Abstract:** A series of graphene-TiO<sub>2</sub> composites was fabricated from graphene oxide and titanium n-butoxide (TNB) by an ultrasonic-assisted method. The structure and composition of the nanocomposites were characterized by Raman spectroscopy, BET surface area measurements, X-ray diffraction, transmission electron microscopy, and ultraviolet-visible absorption spectroscopy. The average size of the TiO<sub>2</sub> nanoparticles on the graphene nanosheets was controlled at around 10–15 nm without using surfactant, which is attributed to the pyrolysis and condensation of dissolved TNB into TiO<sub>2</sub> by ultrasonic irradiation. The catalytic activity of the composites under ultrasonic irradiation was determined using a rhodamine B (RhB) solution. The graphene-TiO<sub>2</sub> composites possessed a high specific surface area, which increased the decolorization rate for RhB solution. This is because the graphene and TiO<sub>2</sub> nanoparticles in the composites interact strongly, which enhances the photoelectric conversion of TiO<sub>2</sub> by reducing the recombination of photogenerated electron-hole pairs.

**Keywords:** [grapheme](#), [ultrasonication](#), [sonocatalytic degradation](#), [adsorption](#), [rhodamine B](#)

收稿日期: 2012-02-18; 出版日期: 2012-07-31

引用本文:

ZHU Lei, Trisha GHOSH, Chong-Yeon PARK 等 .Enhanced Sonocatalytic Degradation of Rhodamine B by Graphene-TiO<sub>2</sub> Composites Synthesized by an Ultrasonic-Assisted Method[J] 催化学报, 2012,V33(8): 1276-1283

ZHU Lei, Trisha GHOSH, Chong-Yeon PARK 等 .Enhanced Sonocatalytic Degradation of Rhodamine B by Graphene-TiO<sub>2</sub> Composites Synthesized by an Ultrasonic-Assisted Method[J] Chinese Journal of Catalysis, 2012,V33(8): 1276-1283

链接本文:

[http://www.chxb.cn/CN/10.1016/S1872-2067\(11\)60430-0](http://www.chxb.cn/CN/10.1016/S1872-2067(11)60430-0) 或 <http://www.chxb.cn/CN/Y2012/V33/I8/1276>

## Service

- ▶ 把本文推荐给朋友
- ▶ 加入我的书架
- ▶ 加入引用管理器
- ▶ Email Alert
- ▶ RSS

## 作者相关文章

- ▶ ZHU Lei
- ▶ Trisha GHOSH
- ▶ Chong-Yeon PARK
- ▶ MENG Ze-Da
- ▶ OH Won-Chun

- [1] Chen X, Mao S S. Chem Rev, 2007, 107: 2891
- [2] Burda C, Chen X, Narayanan R, El-Sayed M A. Chem Rev, 2005, 105: 1025
- [3] Li G, Liu C, Liu Y. Appl Surf Sci, 2006, 253: 2481
- [4] Kumar A, Jain A K. J Mol Catal A, 2001, 165: 265
- [5] Oh W C, Zhang F J, Chen M L. J Ind Eng Chem, 2010, 16: 299
- [6] Hoffmann M R, Martin S T, Choi W, Bahnemann D W. Chem Rev, 1995, 95: 69
- [7] Wang W D, Serp P, Kalck P, Faria J L. J Mol Catal A, 2005, 235: 194

- [8] Woan K, Pyrgiotakis G, Sigmund W. *Adv Mater*, 2009, 21: 2233
- [9] Geim A K, Novoselov K S. *Nat Mater*, 2007, 6: 183
- [10] Gilje S, Han S, Wang M, Wang K L, Kaner R B. *Nano Lett*, 2007, 7: 3394
- [11] Pasricha R, Gupta S, Srivastava A K. *Small*, 2009, 5: 2253
- [12] Robinson J T, Perkins F K, Snow E S, Wei Z Q, Sheehan P E. *Nano Lett*, 2008, 8: 3137
- [13] Arsat R, Breedon M, Sha?ei M, Spizziri P G, Gilje S, Kaner R B, Kalantar-zadeh K, Wlodarski W. *Chem Phys Lett*, 2009, 467: 344
- [14] Stankovich S, Dikin D A, Dommett G H B, Kohlhaas K M, Zimney E J, Stach E A, Piner R D, Nguyen S T, Ruoff R S. *Nature*, 2006, 442: 282
- [15] Seger B, Kamat P V. *J Phys Chem C*, 2009, 113: 7990
- [16] Akhavan O, Ghaderi E, Esfandiar A. *J Phys Chem B*, 2011, 115: 6279
- [17] Liang Y Y, Wang H L, Casalongue H S, Chen Z, Dai H J. *Nano Res*, 2010, 3: 701
- [18] Williams G, Seger B, Kamat P V. *ACS Nano*, 2008, 2: 1487
- [19] Wang D H, Choi D W, Li J, Yang Z G, Nie Z M, Kou R, Hu D H, Wang C M, Saraf L V, Zhang J G, Aksay I A, Liu J. *ACS Nano*, 2009, 3: 907
- [20] Chen C, Cai W M, Long M C, Zhou B X, Wu Y H, Wu D Y, Feng Y J. *ACS Nano*, 2010, 4: 6425
- [21] Zhang H, Lv X J, Li Y M, Wang Y, Li J H. *ACS Nano*, 2010, 4: 380
- [22] Zhang Y H, Tang Z R, Fu X Z, Xu Y J. *ACS Nano*, 2010, 4: 7303
- [23] Nakui H, Okitsu K, Maeda Y, Nishimura R. *J Hazard Mater*, 2007, 146: 636
- [24] Abdullah A Z, Ling P Y. *J Hazard Mater*, 2010, 173: 159
- [25] Wang J, Sun W, Zhang Z H, Jiang Z, Wang X F, Xu R, Li R H, Zhang X D. *J Colloid Interface Sci*, 2008, 320: 202
- [26] Wang J, Lv Y H, Zhang Z H, Deng Y Q, Zhang L Q, Liu B, Xu R, Zhang X D. *J Hazard Mater*, 2009, 170: 398
- [27] Wang J, Sun W, Zhang Z H, Xing Z Q, Xu R, Li R H, Li Y, Zhang X D. *Ultrason Sonochem*, 2008, 15: 301
- [28] Kubo M, Matsuoka K, Takahashi A, Shibasaki-Kitakawa N, Yonemoto T. *Ultrason Sonochem*, 2005, 12: 263
- [29] Wang J, Jiang Z, Zhang L Q, Kang P L, Xie Y P, Lv Y H, Xu R, Zhang X D. *Ultrason Sonochem*, 2009, 16: 225
- [30] Neppolian B, Bruno A, Bianchi C L, Ashokkumar M. *Ultrason Sonochem*, 2012, 19: 9
- [31] Guo J J, Zhu S M, Chen Z X, Li Y, Yu Z Y, Liu Q L, Li J B, Feng C L, Zhang D. *Ultrason Sonochem*, 2011, 18: 1082
- [32] Hummers W S, Offeman R E. *J Am Chem Soc*, 1958, 80: 1339
- [33] Oh W C, Chen M L, Zhang K, Zhang F J, Jang W K, Zhang F J. *J Korean Phys Soc*, 2010, 56: 1097
- [34] Graf D, Molitor F, Ensslin K, Stampfer C, Jungen A, Hierold C, Wirtz L. *Nano Lett*, 2007, 7: 238
- [35] Dresselhaus M S, Jorio A, Hofmann M, Dresselhaus G, Saito R. *Nano Lett*, 2010, 10: 751
- [36] Zhang W X, Cui J C, Tao C A, Wu Y G, Li Z P, Ma L, Wen Y Q, Li G T. *Angew Chem, Int Ed*, 2009, 48: 5864
- [37] Lu J, Yang J X, Wang J, Lim A, Wang S, Loh K P. *ACS Nano*, 2009, 3: 2367
- [38] Akhavan O. *ACS Nano*, 2010, 4: 4174
- [39] Lambert T N, Chavez C A, Hernandez-Sanchez B, Lu P, Bell N S, Ambrosini A, Friedman T, Boyle T J, Wheeler D R, Huber D L. *J Phys Chem C*, 2009, 113: 19812
- [40] Stankovich S, Dikin D A, Piner R D, Kohlhaas KA, Klein-hammes A, Jia Y, Wu Y, Nguyen S T, Ruoff R S. *Carbon*, 2007, 45: 1558
- [41] Qourzal S, Barka N, Tamimi M, Assabbane A, Nounah A, Ihlal A, Ait-Ichou Y. *Mater Sci Eng C*, 2009, 29: 1616
- [42] Gotoh K, Kinumoto T, Fujii E, Yamamoto A, Hashimoto H, Ohkubo T, Itsudani A, Kuroda Y, Ishida H. *Carbon*, 2011, 49: 1118
- [43] Zhang X Y, Li H P, Cui X L, Lin Y H. *J Mater Chem*, 2010, 20: 2801
- [44] Zhang H, Lv X J, Li Y M, Wang Y, Li J H. *ACS Nano*, 2010, 4: 380
- [45] Bourlinos A B, Gournis D, Petridis D, Szabo T, Szeri A, Dekany I. *Langmuir*, 2003, 19: 6050
- [46] Zhu L, Meng Z D, Oh W C. *Chin J Catal*, 2011, 32: 926
- [47] Chen M L, Zhang F J, Oh W C. *New Carbon Mater*, 2009, 24: 159
- [48] Stankovich S, Piner R D, Chen X Q, Wu N Q, Nguyen S T, Ruoff R S. *J Mater Chem*, 2006, 16: 155
- [49] Okitsu K, Iwasaki K, Yobiko Y, Bandow H, Nishimura R, Maeda Y. *Ultrason Sonochem*, 2005, 12: 255
- [50] Shimizu N, Ogino C, Dadjour M F, Murata T. *Ultrason Sono-chem*, 2007, 14: 184
- [51] Zhang K, Oh W C. *Bull Korean Chem Soc*, 2010, 31: 1589
- [52] Saieen J, Delavari H, Solymani A R. *J Hazard Mater*, 2010, 177: 1031
- [53] Wang N, Zhu L H, Wang M Q, Wang D L, Tang H Q. *Ultrason Sonochem*, 2010, 17: 78

- [54] Zhang X W, Zhou M H, Lei L C. Carbon, 2005, 43: 1700
- [55] Liu Z, Robinson J T, Sun X M, Dai H J. J Am Chem Soc, 2008, 130: 10876
- [56] Lightcap I V, Kosel T H, Kamat P V. Nano Lett, 2010, 10: 577
- [57] Akhavan O, Ghaderi E. J Phys Chem C, 2009, 113: 20214
- [1] OH Won-Chun, CHEN Mingliang, CHO Kwangyoun, KIM Cheolkyu, MENG Zeda, ZHU Lei. Synthesis of Graphene-CdSe Composite by a Simple Hydrothermal Method and Its Photocatalytic Degradation of Organic Dyes[J]. 催化学报, 2011, 32(10): 1577-1583
- [2] ZHANG Kan; MENG Zeda; OH Wonchun\*. 紫外光照射下 Fe-碳纳米管/TiO<sub>2</sub> 复合材料降解罗丹明 B[J]. 催化学报, 2010, 31(7): 751-758