催化学报 » 2011, Vol. 32 » Issue (5):736-745 DOI: 10.1016/S1872-2067(10)60230-6

研究论文

最新目录 | 下期目录 | 过刊浏览 | 高级检索

◀◀ 前一篇 后一篇 ▶▶

NOx Reduction on Fully Formulated Lean NOx Trap Catalysts Subjected to Simulated Road Aging: Insights from Steady-State Experiments

Jin WANG, Yaying JI, Uschi GRAHAM, Caio CESAR SPINDOLA DE OLIVEIRA, Mark CROCKER*

Center for Applied Energy Research, University of Kentucky, Lexington, KY 40511, USA

Jin WANG, Yaying JI, Uschi GRAHAM, Caio CESAR SPINDOLA DE OLIVEIRA, Mark CROCKER*

Center for Applied Energy Research, University of Kentucky, Lexington, KY 40511, USA

- 摘要
- 参考文献
- 相关文章

Download: PDF (740KB) HTML (1KB) Export: BibTeX or EndNote (RIS)

摘要 Fully formulated lean NO $_{\chi}$ trap (LNT) catalysts of the type Pt/Rh/BaO/Al $_2$ O $_3$ were prepared with and without incorporation of CeO_2 - ZrO_2 in the washcoat, and their NO_X reduction behavior was evaluated in steady-state, continuous flow experiments. In the fresh state, CeO_2 -Zr O_2 addition was found to exert little effect on NO_X reduction activity using H_2 , CO, and NH_3 as the reductants. However, after simulated road aging, NO_x reduction activity was significantly impaired for the CeO₂-ZrO₂-free catalyst, whereas the performance of the CeO₂-ZrO₂containing analog was affected to only a minor degree. These differences are explained on the basis of highresolution transmission electron microscopy measurements showing that Pt supported on CeO₂-ZrO₂ remained highly dispersed after aging, whereas Pt supported on BaO/Al₂O₃ underwent significant sintering. In addition, the Pt/CeO₂-ZrO₂ component did not accumulate sulfur during aging, unlike Pt/BaO/Al₂O₃ for which significant sulfation of the Ba phase occurred. For both catalysts, selectivity to $\mathrm{NH_3}$ in NO and $\mathrm{NO_2}$ reduction by $\mathrm{H_2}$ increased after catalyst aging, indicative of a change in the relative surface coverages of N and H ad-atoms on the precious metal sites.

关键词: lean NOx trap NOx storage-reduction catalyst steady state NOx reduction ceria-zirconia catalyst aging

Abstract: Fully formulated lean NO_x trap (LNT) catalysts of the type Pt/Rh/BaO/Al₂O₃ were prepared with and without incorporation of CeO_2 - ZrO_2 in the washcoat, and their NO_X reduction behavior was evaluated in steadystate, continuous flow experiments. In the fresh state, ${
m CeO}_2 ext{-}{
m ZrO}_2$ addition was found to exert little effect on ${
m NO}_X$ reduction activity using H_2 , CO, and NH_3 as the reductants. However, after simulated road aging, NO_x reduction activity was significantly impaired for the CeO₂-ZrO₂-free catalyst, whereas the performance of the CeO₂-ZrO₂containing analog was affected to only a minor degree. These differences are explained on the basis of highresolution transmission electron microscopy measurements showing that Pt supported on CeO2-ZrO2 remained highly dispersed after aging, whereas Pt supported on BaO/Al₂O₃ underwent significant sintering. In addition, the Pt/CeO2-ZrO2 component did not accumulate sulfur during aging, unlike Pt/BaO/Al2O3 for which significant sulfation of the Ba phase occurred. For both catalysts, selectivity to NH₃ in NO and NO₂ reduction by H₂ increased after catalyst aging, indicative of a change in the relative surface coverages of N and H ad-atoms on the precious metal sites.

Keywords: lean NOx trap, NOx storage-reduction catalyst, steady state, NOx reduction, ceria-zirconia, catalyst aging

收稿日期: 2011-02-28; 出版日期: 2011-05-03

引用本文:

Jin WANG, Yaying JI, Uschi GRAHAM等 .NOx Reduction on Fully Formulated Lean NOx Trap Catalysts Subjected to Simulated Road Aging: Insights from Steady-State Experiments[J] 催化学报, 2011,V32(5): 736-745

Jin WANG, Yaying JI, Uschi GRAHAM etc .NOx Reduction on Fully Formulated Lean NOx Trap Catalysts Subjected to Simulated Road Aging: Insights from Steady-State Experiments[J] Chinese Journal of Catalysis, 2011, V32(5): 736-745 链接本文:

http://www.chxb.cn/CN/10.1016/S1872-2067(10)60230-6 http://www.chxb.cn/CN/Y2011/V32/I5/736

- eck R M, Farrauto R J. Catalytic Air Pollution Control. 2nd Ed. New York: Wiley, 2002. 119, 206
- oy S, Baiker A. Chem Rev, 2009, 109: 4054
- pling W S, Campbell L E, Yezerets A, Currier N W, Parks J E. Catal Rev-Sci Eng, 2004, 46: 163
- i Y, Choi J-S, Toops T J, Crocker M, Naseri M. Catal Today, 2008, 136: 146

Service

- ▶ 把本文推荐给朋友
- 加入我的书架
- ▶ 加入引用管理器
- ▶ Email Alert
- **▶** RSS

作者相关文章

- Jin WANG
- Yaying JI
- Uschi GRAHAM
- Caio CESAR SPINDOLA DE **OLI VEI RA**
- Mark CROCKER

- 5] i Y, Fisk C, Easterling V, Graham U, Poole A, Crocker M, Choi J-S, Partridge W, Wilson K. Catal Today, 2010, 151: 362 🛺
- [6] aneda M, Morita T, Nagao Y, Kintaichi Y, Hamada H. Phys Chem Chem Phys, 2001, 3: 4696
- [7] i Y Y, Toops T J, Crocker M. Catal Lett, 2007, 119: 257
- [8] hilipp S, Drochner A, Kunert J, Vogel H, Theis J, Lox E S. Top Catal, 2004, 30-31: 235
- [9] ymalla M O, Drochner A, Vogel H, Philipp S, Göbel U, Müller W. Top Catal, 2007, 42-43: 199 and
- [10] Rohart E, Bellière-Baca V, Yokota K, Harlé V, Pitois C. Top Catal, 2007, 42-43: 71
- [11] Phatak A A, Koryabkina N, Rai S, Ratts J L, Ruettinger W, Farrauto R J, Blau G E, Deglass W N, Ribeiro F H. Catal To-day, 2007, 123: 224
- [12] Theis J, Ura J, Goralski C Jr, Jen H, Thanasiu E, Graves Y, Takami A, Yamada H, Miyoshi S. SAE Technical Paper, 2003: 2003-01-1160
- [13] Corbos E C, Courtois X, Bion N, Marecot P, Duprez D. Appl Catal B, 2008, 80: 62
- [14] Kwak J H, Kim D H, Szanyi J, Peden C H F. Appl Catal B, 2008, 84: 545
- [15] Ji Y Y, Toops T J, Crocker M. Catal Lett, 2009, 127: 55
- [16] Easterling V, Ji Y Y, Crocker M, Ura J, Theis J R, McCabe R W. Catal Today, 2010, 151: 338
- [17] Piacentini M, Maciejewski M, Baiker A. Appl Catal B, 2007, 72: 105
- [18] Nagai Y, Hirabayashi T, Dohmae K, Takagi N, Minami T, Shinjoh H, Matsumoto S. J Catal, 2006, 242: 103 🚉
- [19] Eberhardt M, Riedel R, Göbel U, Theis J, Lox E S. Top Catal, 2004, 30-31: 135
- [20] Corbos E C, Elbouazzaoui S, Cortois X, Bion N, Marecot P, Duprez D. Top Catal, 2007, 42-43: 9
- [21] Choi J-S, Partridge W P, Pihl J A, Daw C S. Catal Today, 2008, 136: 173
- [22] Wang J, Ji Y Y, Easterling V, Crocker M, Dearth M, McCabe R W. Catal Today, 2011, doi:10.1016/j.cattod.2011.02.048
- [23] Pihl J A, Parks J E, Daw C S, Root T W. SAE Technical Pa-per, 2006: 2006-01-3441
- [24] Mulla S S, Chaugule S S, Yezerets A, Currier N W, Delgass W N, Ribeiro F H. Catal Today, 2008, 136: 136 and
- [25] Graham G W, Jen H W, Chun W, McCabe R W. J Catal, 1999, 182: 228
- [26] Graham G W, Jen H W, Chun W, Sun H P, Pan X Q, McCabe R W. Catal Lett, 2004, 93: 129
- [27] Xu J, Clayton R D, Balakotaiah V, Harold M P. Appl Catal B, 2008, 77: 395
- [28] Clayton R D, Harold M P, Balakotaiah V. Appl Catal B, 2008, 81: 161
- [29] Siera J, Nieuwenhuys B E, Hirano H, Yamada T, Tanaka K I. Catal Lett, 1989, 3: 179
- [30] Ji Y, Toops T J, Pihl J A, Crocker M. Appl Catal B, 2009, 91: 329
- [31] Hirano H, Yamada T, Tanaka K I, Siera J, Nieuwenhuys B E. Stud Surf Sci Catal, 1993, 75: 345 📠
- [32] Szailer T, Kwak J H, Kim D H, Hanson J C, Peden C H F, Szanyi J. J Catal, 2006, 239: 51
- [33] Lesage T, Verrier C, Bazin P, Saussey J, Daturi M. Phys Chem Chem Phys, 2003, 5: 4435
- [34] Nova I, Lietti L, Forzatti P, Prinetto F, Ghiotti G. Catal Today, 2010, 151: 330
- [35] Martínez-Arias A, Fernández-García M, Iglesias-Juez A, Hungria A B, Anderson J A, Conesa J C, Soria J. Appl Catal B, 2002, 38: 151
- [36] Verdier S, Rohart E, Bradshaw H, Harris D, Bichon Ph, Dela-hay G. SAE Technical Paper, 2008: 2008-01-1022
- [37] Li Y, Cheng H, Li D Y, Qin Y Sh, Wang Sh D. Chin J Catal, 2008, 26: 547
- [38] Pirug G, Bonzel H P. J Catal, 1977, 50: 64
- [39] Clayton R D, Harold M P, Balakotaiah V, Wan C Z. Appl Catal B, 2009, 90: 662
- [40] Bhatia D, Harold M P, Balakotaiah V. Catal Today, 2010, 151: 314
- [41] Xu L, McCabe R, Ruona W, Cavataio G. SAE Technical Paper, 2009: 2009-01-0285

没有找到本文相关文献

