

三维花状 Co_3O_4 的低成本制备及其在催化 CO 氧化中的应用

曹昌燕, 窦智峰, 刘华, 宋卫国*

中国科学院化学研究所, 北京分子科学国家实验室, 分子纳米结构与纳米技术实验室, 北京 100190

CAO Changyan, DOU Zhifeng, LIU Hua, SONG Weiguo*

Beijing National Laboratory for Molecular Science (BNLMS), Laboratory of Molecular Nanostructures and Nanotechnology, Institute of Chemistry, Chinese Academy of Sciences, Beijing 100190, China

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摘要 采用一种快速、无模板、低成本的微波辅助水热法在 2 min 内制备了三维花状 Co_3O_4 。所用原料均是无机盐, 前驱体浓度和尿素的逐渐水解对 Co_3O_4 形貌影响很大。制得的花状 Co_3O_4 比表面积大, 且暴露了 (110) 高活性指数面, 对 CO 氧化具有较高的催化活性。

关键词: 一氧化碳 氧化 四氧化三钴 纳米结构 微波

Abstract: 3D flowerlike Co_3O_4 nanostructures were prepared by a microwave-assisted hydrothermal method, which is a rapid, template-free, and low cost method. The product is obtained in two minutes using all inorganic precursors. The precursor concentration and gradual hydrolysis of urea determine the morphology of Co_3O_4 nanostructures. These flowerlike Co_3O_4 nanostructures have high surface area and expose largely active (110) planes, leading to relative high catalytic activity in CO oxidation.

Keywords: carbon monoxide, oxidation, cobaltosic oxide, nanostructure, microwave

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- [1] Hu L H, Peng Q, Li Y D. J Am Chem Soc, 2008, 130: 16136
- [2] Xie X W, Li Y, Liu Z Q, Haruta M, Shen W J. Nature, 2009, 458: 746
- [3] Xie X W, Shang P J, Liu Z Q, Lv Y G, Li Y, Shen W J. J Phys Chem C, 2010, 114: 2116
- [4] Xie R Y, Li D B, Hou B, Wang J A, Jia L T, Sun Y H. Catal Commun, 2010, 12: 380
- [5] Cao A M, Hu J S, Liang H P, Song W G, Wan L J, He X L, Gao X G, Xia S H. J Phys Chem B, 2006, 110: 15858
- [6] Wang X, Wu X L, Guo Y G, Zhong Y T, Cao X Q, Ma Y, Yao J N. Adv Funct Mater, 2010, 20: 1680
- [7] Li W Y, Xu L N, Chen J. Adv Funct Mater, 2005, 15: 851
- [8] Barakat M N A, Khil M S, Sheikh F A, Kim H Y. J Phys Chem C, 2008, 112: 12225
- [9] Wang G X, Shen X P, Horvat J, Wang B, Liu H, Wexler D, Yao J. J Phys Chem C, 2009, 113: 4357
- [10] Jansson J, Palmqvist A E C, Fridell E, Skoglundh M, Osterlund L, Thormahlen P, Langer V. J Catal, 2002, 211: 387
- [11] Yu Y, Takei T, Ohashi H, He H, Zhang X, Haruta M. J Catal, 2009, 267: 121
- [12] Benjaramtg M R, Katta L. Chin J Catal (催化学报), 2011, 32: 800
- [13] Kim M H, Kim D W, Gode T. Chin J Catal (催化学报), 2011, 32: 762
- [14] Qiao B, Wang A, Lin J, Li L, Su D, Zhang T. Appl Catal B, 2011, 105: 103
- [15] Hu L H, Sun K Q, Peng Q, Xu B Q, Li Y D. Nano Res, 2010, 3: 363

- [16] Tian B Z, Liu X Y, Yang H F, Xie S H, Yu C Z, Tu B , Zhao D Y. *Adv Mater*, 2003, 15: 1370
- [17] Yang L X, Zhu Y J, Li L, Zhang L, Tong H, Wang W W, Cheng G F, Zhu J F. *Eur J Inorg Chem*, 2006, 2006: 4787
- [18] Zhao Z G, Geng F X, Bai J B, Cheng H M. *J Phys Chem C*, 2007, 111: 3848
- [19] Qiao R, Zhang X L, Qiu R, Kim J C, Kang Y S. *Chem Eur J*, 2009, 15: 1886
- [20] Zheng J, Liu J, Lv D P, Kuang Q, Jiang Z Y, Xie Z X, Huang R B, Zheng L S. *J Solid State Chem*, 2010, 183: 600
- [21] Chen A, Peng X, Koczkur K, Miller B. *Chem Commun*, 2004: 1964 
- [22] Zhang H, Yang D, Ji Y J, Ma X Y, Xu J, Que D L. *J Phys Chem B*, 2004, 108: 3955
- [23] Sun C W, Sun J, Xiao G L, Zhang H R, Qiu X P, Li H, Chen L Q. *J Phys Chem B*, 2006, 110: 13445
- [24] Zhong L S, Hu J S, Liang H P, Cao A M, Song W G, Wan L J. *Adv Mater*, 2006, 18: 2426
- [25] Zhong L S, Hu J S, Cao A M, Liu Q, Song W G, Wan L J. *Chem Mater*, 2007, 19: 1648
- [26] Bain S W, Ma Z, Cui Z M, Zhang L S, Niu F, Song W G, Wan L J. *J Phys Chem C*, 2008, 112: 11340
- [27] Cao C Y, Cui Z M, Chen C Q, Song W G, Cai W. *J Phys Chem C*, 2010, 114: 9865
- [28] Cao C Y, Guo W, Cui Z M, Song W G, Cai W. *J Mater Chem*, 2011, 21: 3204
- [29] Zeng S, Tang K, Li T, Liang Z, Wang D, Wang Y, Qi Y, Zhou W. *J Phys Chem C*, 2008, 112: 4836
- [30] Long M, Cai W M, Cai J, Zhou B X, Chai X Y, Wu Y H. *J Phys Chem B*, 2006, 110: 20211
- [31] Guttel R, Paul M, Schuth F. *Chem Commun*, 2010, 46: 895
- [1] 张元卓, 于兹瀛, 张富民, 肖强, 钟依均, 朱伟东. 纳米 Li_2ZrO_3 吸收剂原位移除 CO_2 强化水煤气变换反应制氢[J]. 催化学报, 2012, 33(9): 1572-1577
- [2] 贾燕子, 杨清河, 孙淑玲, 聂红, 李大东. 汽油加氢处理过程中 $\text{Mo-V}/\text{Al}_2\text{O}_3$ 的催化性能及协同效应[J]. 催化学报, 2012, 33(9): 1546-1551
- [3] 石川, 徐力, 朱爱民, 张玉卓. 区择择氧化铈稳定的 CuO 簇在 $\text{CO}, \text{C}_3\text{H}_6$ 和 NO 消除中的催化性能[J]. 催化学报, 2012, 33(9): 1455-1462
- [4] 司维峰, 李焕巧, 尹杰, 李书双, 谢妍, 李佳, 吕洋, 刘元, 邢永恒, 徐缓, 宋玉江. 球形分枝结构 Pt 纳米催化剂的合成、纯化及电催化性能[J]. 催化学报, 2012, 33(9): 1601-1607
- [5] 洪伟, 刘百军, 王宏宾, 陈玉. $\text{TiO}_2-\text{Al}_2\text{O}_3$ 的水热法合成及其负载的 NiMoP 催化剂上 FCC 柴油加氢脱硫性能[J]. 催化学报, 2012, 33(9): 1586-1593