

Sr 取代 LaFeO₃ 钙钛矿的结构性质和催化性能

张晓静, 李华举, 李勇, 申文杰*

中国科学院大连化学物理研究所催化基础国家重点实验室, 辽宁大连 116023

ZHANG Xiaojing, LI Huaju, LI Yong, SHEN Wenjie*

State Key Laboratory of Catalysis, Dalian Institute of Chemical Physics, Chinese Academy of Sciences, Dalian 116023, Liaoning, China

- 摘要
- 参考文献
- 相关文章

Download: PDF (877KB) HTML (1KB) Export: BibTeX or EndNote (RIS) Supporting Info

摘要 Sr²⁺对La³⁺的部分取代导致LaFeO₃的结构性质和催化性能发生了显著变化。钙钛矿结构由LaFeO₃的正交型变成了La_{0.8}Sr_{0.2}FeO₃的近立方型。由于电荷补偿效应, Sr²⁺取代La³⁺导致部分Fe³⁺氧化为Fe⁴⁺, 同时产生氧空穴, 因而提高了La_{0.8}Sr_{0.2}FeO₃的还原性能。由于氧空穴的作用, La_{0.8}Sr_{0.2}FeO₃催化剂在CO氧化和CH₄燃烧反应中均表现出较LaFeO₃高的催化活性。在CO氧化反应中, 氧空穴有利于反应物分子的吸附并加速了气相氧分子在表面上的解离; 而在CH₄燃烧反应中, 氧空穴则促进了晶格氧物种从体相到表面的扩散。

关键词: 钙钛矿 铁酸镧 取代 还原性 一氧化碳氧化 甲烷燃烧

Abstract: Partial substitution of La³⁺ by Sr²⁺ in LaFeO₃ resulted in significant changes in its structure and catalytic activity. The perovskite structure was changed from orthorhombic in LaFeO₃ to nearly cubic in La_{0.8}Sr_{0.2}FeO₃. Replacement of La³⁺ by Sr²⁺ induced a positive charge deficiency that was compensated for by the oxidation of some Fe³⁺ to Fe⁴⁺ and the generation of oxygen vacancies, which greatly promoted the reducibility of the perovskite. La_{0.8}Sr_{0.2}FeO₃ gave considerably enhanced activity in CO oxidation and methane combustion because the oxygen vacancies accelerated the dissociation of gaseous oxygen on the surface in CO oxidation and facilitated the diffusion of lattice oxygen from the bulk to the surface during CH₄ combustion.










Keywords: perovskite, lanthanum ferrite, substitution, reducibility, carbon monoxide oxidation, methane combustion

收稿日期: 2012-02-14; 出版日期: 2012-04-23

引用本文:
张晓静, 李华举, 李勇等. Sr 取代 LaFeO₃ 钙钛矿的结构性质和催化性能[J] 催化学报, 2012, V33(7): 1109-1114

ZHANG Xiao-Jing, LI Hua-Ju, LI Yong etc. Structural Properties and Catalytic Activity of Sr-Substituted LaFeO₃ Perovskite[J] Chinese Journal of Catalysis, 2012, V33(7): 1109-1114

链接本文:
http://www.chxb.cn/CN/10.1016/S1872-2067(11)60388-4 或 http://www.chxb.cn/CN/Y2012/V33/I7/1109

- [1] Ziaei-Azad H, Khodadadi A, Esmailnejad-Ahranjani P, Mor-tazavi Y. Appl Catal B, 2011, 102: 62 
- [2] Tejuca L G, Fierro J L G. Properties and Applications of Perovskite-Type Oxides. New York: Marcel Dekker, 1993. 1
- [3] Ciambelli P, Cimino S, Lisi L, Faticanti M, Minelli G, Pettiti I, Porta P. Appl Catal B, 2001, 33: 193 
- [4] Tanaka H, Taniguchi M, Kajita N, Uenishi M, Tan I, Sato N, Narita K, Kimura M. Top Catal, 2004, 30/31: 389 
- [5] Tan I, Tanaka H, Uenishi M, Kaneko K, Mitachi S. J Ceram Soc Jpn, 2005, 113: 71 
- [6] Wang Y G, Ren J W, Wang Y Q, Zhang F Y, Liu X H, Guo Y, Lu G Zh. J Phys Chem C, 2008, 112: 15293 
- [7] Najjar H, Lamonier J F, Mentre O, Giraudon J M, Batis H. Appl Catal B, 2011, 106: 149
- [8] Tanaka H, Misono M. Curr Opin Solid State Mater Sci, 2001, 5: 381 
- [9] Tanaka H, Taniguchi M, Uenishi M, Kajita N, Tan I, Nishihata Y, Mizuki J, Narita K, Kimura M, Kaneko K. Angew Chem, Int Ed, 2006, 45: 5 
- [10] Nishihata Y, Mizuki J, Akao T, Tanaka H, Uenishi M, Kimura M, Okamoto T, Hamada N. Nature, 2002, 418: 164 
- [11] Uenishi M, Tanaka H, Taniguchi M, Tan I, Nishihata Y, Mizuki J, Kobayashi T. Catal Commun, 2008, 9: 311 

Service

- ▶ 把本文推荐给朋友
- ▶ 加入我的书架
- ▶ 加入引用管理器
- ▶ Email Alert
- ▶ RSS

作者相关文章

- ▶ 张晓静
- ▶ 李华举
- ▶ 李勇
- ▶ 申文杰

- [12] Matsumura D, Nishihata Y, Mizuki J, Taniguchi M, Uenishi M, Tanaka H. *J Appl Phys*, 2010, 107: 124319 
- [13] Leanza R, Rossetti I, Fabbrini L, Oliva C, Forni L. *Appl Catal B*, 2000, 28: 55 
- [14] Merino N A, Barbero B P, Grange P, Cadus L E. *J Catal*, 2005, 231: 232 
- [15] Pecchi G, Reyes P, Zamora R, Campos C, Cadus L E, Barbero B P. *Catal Today*, 2008, 133-135: 420 
- [16] Deng J G, Dai H X, Jiang H Y, Zhang L, Wang G Zh, He H, Au C T. *Environ Sci Technol*, 2010, 44: 2618 
- [17] Wang Ch H, Chen Ch L, Weng H Sh. *Chemosphere*, 2004, 57: 1131 
- [18] Ferri D, Forni L. *Appl Catal B*, 1998, 16: 119 
- [19] Ponce S, Pena M A, Fierro J L G. *Appl Catal B*, 2000, 24: 193 
- [20] Ciambelli P, Cimino S, De Rossi S, Lisi L, Minelli G, Porta P, Russo G. *Appl Catal B*, 2001, 29: 239 
- [21] Zhang X J, Li H J, Li Y, Shen W J. *Catal Lett*, 2012, 142: 118 
- [22] Zhang X J, Li Y, Li H J, Shen W J. *J Nat Gas Chem*, 2012, 21: 113 
- [23] Cullity B D. *Elements of X-Ray Diffraction*. 2nd Ed. Menlo Park: Addison-Wesley Press, 1978. 102 
- [24] Barbero B P, Gamboa J A, Cadus L E. *Appl Catal B*, 2006, 65: 21 
- [25] Pecchi G, Jiliberto M G, Delgado E J, Cadus L E, Fierro J L G. *J Chem Technol Biotechnol*, 2011, 86: 1067 
- [26] Shannon R D. *Acta Crystallogr A*, 1976, A32: 751
- [27] Ciambelli P, Cimino S, De Rossi S, Faticanti M, Lisi L, Minelli G, Pettiti I, Porta P, Russo G, Turco M. *Appl Catal B*, 2000, 24: 243 
- [28] Zhang R D, Villanueva A, Alamdari H, Kaliaguine S. *J Catal*, 2006, 237: 368 
- [29] Voorhoeve R J H, Remeika J P, Trimble L E. *Ann N Y Acad Sci*, 1976, 272: 3 
- [30] Teng F, Han W, Liang S H, Gaugeu B, Zong R L, Zhu Y F. *J Catal*, 2007, 250: 1 
- [31] Chen Ch Q, Li W, Cao Ch Y, Song W G. *J Mater Chem*, 2010, 20: 6968 
- [1] 孙明娟, 邹国军, 许珊, 王晓来. 前驱体 $\text{Ce}(\text{OHCO}_3)_2$ 的结构对产物 CeO_2 催化性能的影响[J]. *催化学报*, 2012,33(8): 1318-1325
- [2] 张慧丽, 任丽会, 陆安慧, 李文翠. $\text{Au}/\text{CeO}_2/\text{SiO}_2$ 催化CO低温氧化反应过程中 CeO_2 的作用[J]. *催化学报*, 2012,33(7): 1125-1132
- [3] 唐富顺, 庄柯, 杨芳, 杨利利, 许波连, 邱金恒, 范以宁. 负载型 $\text{V}_2\text{O}_5/\text{TiO}_2$ 催化剂表面分散状态和性质对氨选择性催化还原 NO 性能的影响[J]. *催化学报*, 2012,33(6): 933-940
- [4] 郭荷芹, 李德宝, 陈从标, 范志宏, 孙予罕. $\text{V}_2\text{O}_5/\text{CeO}_2$ 催化剂上甲醇氧化一步法合成二甲氧基甲烷[J]. *催化学报*, 2012,33(5): 813-818
- [5] 郝芳, 钟俊, 刘平乐, 游奎一, 魏超, 罗和安. 金属取代型 AIPO-5 分子筛催化剂上环己烷亚硝化一步法合成己内酰胺[J]. *催化学报*, 2012,33(4): 670-676