

# BaTiO<sub>3</sub> 纳米颗粒的聚丙烯酰胺凝胶法合成及光催化降解甲基红性能

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**摘要** 采用聚丙烯酰胺凝胶法合成了 BaTiO<sub>3</sub> 纳米颗粒, 利用 X 射线衍射、傅里叶变换红外光谱、透射电镜和紫外-可见漫反射光谱对样品进行了表征。结果表明, 以柠檬酸为络合剂、pH = 2 且在 700 °C 烧烧时可制备出单相 BaTiO<sub>3</sub> 纳米颗粒, 其形状较为规整, 近似呈球形, 平均粒径约为 55 nm, 光学带隙值为 3.25 eV。以偶氮染料甲基红为目标降解物, 研究了 BaTiO<sub>3</sub> 纳米颗粒的光催化性能。结果表明, 在紫外光照射下该纳米颗粒表现出较高的催化活性, 光催化机理主要为光生空穴的直接氧化。

**关键词:** 钛酸钡 纳米颗粒 丙烯酰胺凝胶法 光催化 甲基红

**Abstract:** A polyacrylamide gel method was used to synthesize BaTiO<sub>3</sub> nanoparticles. X-ray diffraction, Fourier transform infrared spectroscopy, transmission electron microscopy, and ultraviolet-visible diffuse reflectance spectroscopy were adopted to characterize the sample. The results demonstrate that single-phase BaTiO<sub>3</sub> nanoparticles can be prepared at a calcination temperature of 700 °C when using citric acid as the chelating agent at pH = 2. The prepared particles are regularly shaped like spheres with an average particle size of ~55 nm and have an optical bandgap energy of 3.25 eV. The photocatalytic properties of BaTiO<sub>3</sub> nanoparticles were investigated using the degradation of the azo dye methyl red. The experimental results reveal that the nanoparticles exhibit a pronounced photocatalytic activity for the methyl red degradation under ultraviolet light irradiation, and the direct oxidation by the hole is suggested to be the main mechanism responsible for the dye decomposition.

**Keywords:** barium titanate, nanoparticle, polyacrylamide gel method, photocatalysis, methyl red

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- [1] Puangpatch T, Sommakettarin P, Chavadej S, Sreethawong T. Int J Hydrogen Energ, 2010, 35: 12428
- [2] Sulaeman U, Yin S, Sato T. Appl Catal B, 2011, 102: 286
- [3] 县涛, 杨华, 戴剑锋, 魏智强, 马金元, 冯旺军. 催化学报 (Xian T, Yang H, Dai J F, Wei Zh Q, Ma J Y, Feng W J. Chin J Catal), 2011, 32: 618
- [4] 王仕发, 杨华, 县涛. 催化学报 (Wang Sh F, Yang H, Xiao T. Chin J Catal), 2011, 32: 1199
- [5] Dong B, Li Z C, Li Z Y, Xu X R, Song M X, Zheng W, Wang C, Al-Deyab S S, El-Newehy M. J Am Ceram Soc, 2010, 93: 3587
- [6] Karaki T, Yan K, Miyamoto T, Adach M. Jpn J Appl Phys, 2007, 46: L97
- [7] Sharma H B, Sarma H N K, Mansingh A. J Mater Sci, 1999, 34: 1385
- [8] Lang S B, Rice L H, Shaw S A. J Appl Phys, 1969, 40: 4335
- [9] Simon-Seveyrat L, Hajjaji A, Emziane Y, Guiffard B, Guyomar D. Ceram Inter, 2007, 33: 35
- [10] Sydorchuk V, Khalameida S, Zazhigalov V. Annales UMCS, 2009, 11: 159
- [11] Chen Y H, Chen Y D. J Hazard Mater, 2011, 185: 168

- [12] 牛新书, 陈晓丽, 李华, 茹祥莉, 李自强. 化学研究与应用 (Niu X Sh, Chen X L, Li H, Ru X L, Li Z Q. Chem Res Appl), 2009, 21: 1296
- [13] GomathiDevi L, Krishnamurthy G. J Hazard Mater, 2009, 162: 899 
- [14] GomathiDevi L, Krishnamurthy G. J Phys Chem A, 2011, 115: 460 
- [15] Kavian R, Saidi A. J Alloys Compd, 2009, 468: 528 
- [16] 陈妍妍, 张云, 王晓燕. 化学学报 (Chen Y Y, Zhang Y, Wang X Y. Acta Chim Sin), 2010, 68: 2409
- [17] Xue L H, Yan Y W. J Nanosci Nanotechnol, 2010, 10: 973 
- [18] Suzuki K, Tanaka N, Kageyama K, Takagi H. J Mater Res, 2009, 24: 1543 
- [19] 丁西亚, 沈波, 翟继卫, 付芳, 张景基, 姚熹. 硅酸盐学报 (Ding X Y, Shen B, Zhai J W, Fu F, Zhang J J, Yao X. J Chin Ceram Soc), 2009, 37: 128
- [20] Terashi Y, Purwanto A, Wang W N, Iskandar F, Okuyama K. J Eur Ceram Soc, 2008, 28: 2573 
- [21] Yang H, Cao Z E, Shen X, Jiang J L, Wei Zh Q, Dai J F, Feng W J. Mater Lett, 2009, 63: 655 
- [22] Zhang Y C, Wang G L, Li K W, Zhang M, Hu X Y, Wang H. J Cryst Growth, 2006, 290: 513 
- [23] Kubelka P, Munk F. Z Tech Phys, 1931, 12: 593
- [24] Tauc J, Grigorovici R, Vancu A. Phys Stat Sol, 1966, 15: 627 
- [25] 崔爱莉, 陈仁政, 尉京志, 李龙土. 无机化学学报 (Cui A L, Chen R Zh, Wei J Zh, Li L T. Chin J Inorg Chem), 2001, 17: 627
- [26] 郭鹏, 刘春燕, 高敏, 王祥生, 郭洪臣. 催化学报 (Guo P, Liu Ch Y, Gao M, Wang X Sh, Guo H Ch. Chin J Catal), 2010, 31: 573
- [27] Ayed L, Mandhi A, Cheref A, Bakhrouf A. Desalination, 2011, 274: 272 
- [28] Cuong D D, Lee J. Integr Ferroelectr, 2006, 84: 23 
- [29] Behnajady M A, Modirshahla N, Shokri M. Chemosphere, 2004, 55: 129 
- [1] 胡全红, 黎先财, 杨爱军, 杨春燕. $\text{BaTiO}_3\text{-BaAl}_2\text{O}_4\text{-Al}_2\text{O}_3$ 复合载体的制备、表征及其Ni基催化剂催化 $\text{CH}_4/\text{CO}_2$ 重整反应性能[J]. 催化学报, 2012, 33(3): 563-569
- [2] 景明俊, 王岩, 钱俊杰, 张敏, 杨建军.水热法制备铂掺杂二氧化钛及其可见光催化性能[J]. 催化学报, 2012, 33(3): 550-556
- [3] 黄燕, 李可心, 颜流水, 戴玉华, 黄智敏, 薛昆鹏, 郭会琴, 熊晶晶.二维六方 $p6mm$ 有序介孔 $\text{WO}_3\text{-TiO}_2$ 复合材料的制备及其可见光光催化性能[J]. 催化学报, 2012, 33(2): 308-316
- [4] 任远航, 辜敏, 胡怡晨, 岳斌, 江磊, 孔祖萍, 贺鹤勇b.稀土负载钛-硅沸石ETS-10的制备及其光催化性质[J]. 催化学报, 2012, 33(1): 123-128
- [5] 王晟, 高艳龙, 王駒, 王栋良, 丁源维, 许学飞, 张晓龙, 江国华.紫外光还原法制备铂填充硅钛复合纳米管及其光催化性能[J]. 催化学报, 2011, 32(9): 1513-1518