

论文

液相还原法制备的纳米晶Ag-50Ni合金在含Cl⁻介质中腐蚀电化学行为研究

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摘要:

用液相还原法制备纳米尺寸Ag-50Ni粉末而后热压制得其纳米块体合金,并与传统粉末冶金法制备的粗晶Ag-50Ni合金对比研究了它们在含Cl⁻介质中的腐蚀电化学行为.结果表明:合金粉末平均粒径约为45nm,真空热压后,晶粒有所长大,但仍为纳米尺度;随Cl⁻浓度增加,两种尺寸Ag-50Ni合金的腐蚀电流密度均增加,腐蚀速度加快;晶粒尺寸降低后,腐蚀电流密度略有增加,腐蚀速度变快;当极化电位增到某一程度后,两种尺寸Ag-50Ni合金均出现钝化.纳米尺寸Ag-50Ni合金在含0.02mol/L Cl⁻介质中的交流阻抗谱由双容抗弧组成,其余均由单容抗弧组成,腐蚀受电化学反应控制.

关键词: 液相还原 Ag-Ni合金 纳米晶 腐蚀电化学

ELECTROCHEMICAL CORROSION BEHAVIOR IN CHLORIDE ION CONTAINING MEDIA OF A NANOCRYSTALLINE BULK Ag-50Ni ALLOY

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Abstract:

Precursory alloy powders with an average particle size of about 45nm were obtained by aqueous reducing method firstly and then a nanocrystalline Ag-50Ni bulk alloy was prepared through hot pressing the precursor at 620°C under a pressure of 58 MPa. The grain sizes of the bulk alloy after hot pressing still remained in nano-size range. The electrochemical corrosion behavior of the nano-alloy was compared in chloride ion containing media with the corresponding coarse grained alloy prepared by conventional powder metallurgy. Corrosion current densities of nanocrystalline and coarse grained Ag

-50Ni alloys increase and therefore corrosion rates become faster with the increment of chloride ion concentrations. However, the corrosion current densities of nanocrystalline Ag-50Ni are a little bigger than those of the corresponding coarse grained Ag-50Ni. Ag-50Ni alloys showed passivation behavior in media containing chloride ions with increment of polarization potentials. EIS plot of nanocrystalline Ag-50Ni is composed of single capacitive loop in a medium containing 0.05 mol/L Na₂SO₄ plus 0.02 mol/L NaCl, while the others are composed of double capacitive loops. Thus, corrosion processes are controlled by electrochemical reactions.

Keywords: aqueous reducing Ag-Ni alloy nanocrystalline corrosion electrochemistry

收稿日期 2008-10-07 修回日期 2009-08-10 网络版发布日期 2009-04-24

DOI:

基金项目:

国家自然科学基金(50771068), 辽宁省自然科学基金(20062049)和辽宁省教育厅创新团队项目(2007T166)

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