

咪唑类离子液体混合物吸收CO₂性能研究王梅^{1,2}, 张立麒¹, 刘浩², 张军营¹, 郑楚光¹

1. 华中科技大学 煤燃烧国家重点实验室, 湖北 武汉 430074;

2. 湖北工业大学 化学与环境工程学院, 湖北 武汉 430068

Studies on CO₂ absorption performance by imidazole-based ionic liquid mixturesWANG Mei^{1,2}, ZHANG Li-qi¹, LIU Hao², ZHANG Jun-ying¹, ZHENG Chu-guang¹

1. State Key Laboratory of Coal Combustion, Huazhong University of Science and Technology, Wuhan 430074, China;

2. School of Chemical and Environmental Engineering, Hubei University of Technology, Wuhan 430068, China

- 摘要
- 参考文献
- 相关文章
- 点击分布统计
- 下载分布统计

全文: [PDF](#) (500 KB) [HTML](#) (1 KB) 输出: [BibTeX](#) | [EndNote](#) (RIS) [背景资料](#)

摘要 结合常规离子液体和功能型离子液体在吸收CO₂方面的优势, 将两类咪唑类离子液体进行混合, 对其吸收CO₂的效果和再生性能进行了实验研究。结果表明, 两类咪唑类离子液体混合后流动性明显改善, 与CO₂接触气液传质顺畅; 常规离子液体[bmim][BF₄]和[bmim][Tf₂N]与胺功能型离子液体[NH₂e-mim][BF₄]混合物较单一的离子液体吸收CO₂的量大, [bmim][CH₃CO₂]与[NH₂e-mim][BF₄]混合后较单一的[bmim][CH₃CO₂]吸收量有明显的减低; 随着常规咪唑类离子液体阳离子碳链增长, 混合离子液体吸收CO₂的效果变强; 与胺乙基功能型离子液体混合吸收CO₂时, 阴离子为[Tf₂N]的常规咪唑类离子液体要比阴离子为[BF₄]的吸收效果好; 离子液体混合物吸收CO₂后经再生循环利用10次, 混合物质量基本不变, 循环使用后吸收CO₂性能为初始吸收性能的75%~85%。

关键词: 咪唑类离子液体 CO₂减排 CO₂吸收 再生性能

Abstract: Conventional and functional imidazole-based ionic liquids (abbr. ILs) were mixed based on their advantage and disadvantage on CO₂ reduction. Additionally, CO₂ absorption effect and regeneration performance of imidazole-based IL mixtures were discussed. It was showed that imidazole-based IL mixtures had good fluidity and smooth of transferring CO₂. It had better absorption capacity of CO₂ for the mixtures of [bmim][BF₄] (or [bmim][Tf₂N]) and [NH₂e-mim][BF₄] than the single IL, and lower absorption capacity for the mixtures of [bmim][CH₃CO₂] and [NH₂e-mim][BF₄] than [bmim][CH₃CO₂]. While the cation of conventional imidazolium ILs became longer and the mixtures could absorb CO₂ more obviously, more strong effect was shown on CO₂ absorption with the anion [Tf₂N] than the anion [BF₄] for the conventional imidazolium IL. CO₂ absorption capacity of the imidazole-based IL mixtures had maintained 75%~85% of the initial capacity during 10 times of the absorption/regeneration cycles, while the quality of the regeneration was unchanged.

Key words: imidazole-based ionic liquids CO₂ emission reduction CO₂ absorption regeneration performance

收稿日期: 2012-04-28;

基金资助:

国家自然科学基金(51076056, 51021065); 国家重点基础研究发展规划(973计划, 2011CB707301); 煤燃烧国家重点实验室开放基金(FSKLCC1111)。

通讯作者: 张立麒, E-mail: lqzhang@mail.hust.edu.cn. E-mail: lqzhang@mail.hust.edu.cn

引用本文:

王梅, 张立麒, 刘浩等. 咪唑类离子液体混合物吸收CO₂性能研究[J]. 燃料化学学报, 2012, 40(10): 1264-1268.

服务

- ▶ 把本文推荐给朋友
- ▶ 加入我的书架
- ▶ 加入引用管理器
- ▶ E-mail Alert
- ▶ RSS

作者相关文章

- ▶ 王梅
- ▶ 张立麒
- ▶ 刘浩
- ▶ 张军营
- ▶ 郑楚光

- [1] 郑楚光. 温室效应及其控制对策[M]. 北京:中国电力出版社,2001: 16-26. (ZHENG Chu-guang. The greenhouse effect and control countermeasures[M]. Beijing: China Power Press, 2001: 16-26.)
- [2] FIGUEROA J D, FOUT T, PLASYNISKI S, MCILVRIED H, SRIVASTAVA R D. Advances in CO₂ capture technology- The U. S. Department of Energy's Carbon Sequestration Program[J]. *J Greenhouse Gas Control*, 2008, 2: 9-20.
- [3] LIM B-H, CHOE W-H, SHIM J-J, RA C S, TUME D, LEE H, LEE C S. High-pressure solubility of carbon dioxide in imidazolium-based ionic liquids with anions and [J]. *Korean J Chem Eng*, 2009, 26(4): 1130-1136.
- [4] 张锁江, 吕兴梅. 离子液体——从基础研究到工业应用[M]. 北京: 科学出版社, 2006: 408-427. (ZHANG Suo-jiang, LV Xing-mei. Ionic liquid—from basic research to industrial application[M]. Beijing: Science Press, 2006: 408-427.)
- [5] BLANCHARDL A, HANCU D, BECKMAN E B, BRENNERKE J F. Green processing using ionic liquids and CO₂[J]. *Nature*, 1999, 399(6731):28-29.
- [6] ZHU X, LU Y-X, PENG C-J, HU J, LIU H-L, HU Y. Halogen bonding interactions between brominated ion pairs and CO₂ molecules: implications for design of new and efficient ionic liquids for CO₂ absorption[J]. *J Phys Chem B*, 2011, 115(14): 3949-3958.
- [7] SHFLETT M B, KASPRZAK D J, JUNK C P, YOKOZEKI. A Phase behavior of {carbon dioxide + } mixtures[J]. *J Chem Thermodynamics*, 2008, 40(1): 25-31.
- [8] HUANG J, RIISAGER A, BERG R W, FEHRMANN R. Tuning ionic liquids for high gas solubility and reversible gas absorption [J]. *J Mol Catal A*, 2008, 297(2): 170-176.
- [9] PENNLINE H W, LUEBKE D R, JONES K L, MYERS C R, MORSI B I, HEINTZ Y J, ILCONICHJ B. Progress in carbon dioxide capture and separation research for gasification-based power generation point sources[J]. *Fuel Process Technol*, 2008, 89(9): 897-907.
- [10] GOODRICH B F, de la FUENTE J C, GURKAN B E, ZADIGIAN D J, PRICE E A, HUANG Y, BRENNERKE J F. Experimental measurements of amine-functionalized anion-tethered ionic liquids with carbon dioxide[J]. *Ind Eng Chem Res*, 2011, 50(1): 111-118.
- [11] BATES E D, MAYTON R D, NTAI I, DAVIS J H, Jr. CO₂ capture by a task-specific ionic liquid[J]. *J Am Chem Soc*, 2002, 124(6): 926-927.
- [12] C HEN J, MA J-T, JI Y, SUN X-Q. Solvent impregnated resin prepared using task-specific ionic liquids for rare earth separation[J]. *J Rare Earths*, 2009, 27(6): 932-936.
- [13] YUE Q-C, FENG L, YAN Q-P, GONG H-S. Preparation and characterization of amino or carboxyl-functionalized ionic liquids[J]. *Chin Chem Lett*, 2007, 18(1): 21-23.
- [14] ANTHONY J L, ANDERSON J L, MAGINN E J, BRENNERKE J F. Anion effects on gas solubility in ionic liquids[J]. *J Phys Chem B*, 2005, 109(13): 6366-6374.
- [15] SCOVAZZO P, CAMPER D, KIEFT J, POSHUSTAJ, KOVAL C, NOBLE R. Regular solution theory and CO₂ gas solubility in room temperature ionic liquids[J]. *Ind Eng Chem Res*, 2004, 43(21): 6855-6860.
- [16] AKI S N V K, MELLEIN B R, SAURER E M, BRENNERKE J F. High-pressure phase behavior of carbon dioxide with imidazolium-based ionic liquids [J]. *J Phys Chem B*, 2004, 108(52): 20355-20365.
- [17] WANG G, HOU W, XIAO F, GENG J, WU Y, ZHANG Z. Low-viscosity triethylbutyl ammonium acetate as a task-specific ionic liquid for reversible CO₂ absorption[J]. *J Chem Eng Data*, 2011, 56(4): 1125-1133.
- [18] SCHREINER C, ZUGMANN S, HARTL R, GORESH J. Fractional walden rule for ionic liquids: Examples from recent measurements and a critique of the so-called ideal KCl line for the walden plot[J]. *J Chem Eng Data*, 2010, 55(5): 1784-1788.
- [19] MULDOON M J, AKI S N V K, ANDERSON J L, DIXON J K, BRENNERKE J F. Improving carbon dioxide solubility in ionic liquids[J]. *J Phys Chem B*, 2007, 111(30): 9001-9009.
- [1] 乔春珍, 王宝利, 肖云汉. 不同钙基CO₂吸收剂的循环特性研究[J]. 燃料化学学报, 2010, 38(04): 478-482.
- [2] 胡恩源, 闫常峰, 蔡炽柳, 胡蓉蓉. 生物油水溶性组分的水蒸气催化重整制氢实验研究[J]. 燃料化学学报, 2009, 37(02): 177-182.

版权所有 © 《燃料化学学报》编辑部

本系统由北京玛格泰克科技发展有限公司设计开发 技术支持: support@magtech.com.cn