

[home](#)[about](#)[publishers](#)[editorial boards](#)[advisory board](#)[for authors](#)[call for papers](#)[subscription](#)[archive](#)[news](#)[links](#)[contacts](#)[authors gateway](#)

Are you an author in Thermal science? In preparation.

THERMAL SCIENCE

International Scientific Journal

[George E. Skodras](#), [Panagiotis Natas](#), [Panagiotis P. Basinas](#), [George P. Sakellariopoulos](#)

REMOVAL OF POLLUTANTS FROM POOR QUALITY COALS BY PYROLYSIS

ABSTRACT

Combustion of poor quality coals and wastes is used today worldwide for energy production. However, this entails significant environmental risks due to the presence of polluting compounds in them, i.e. S, N, Hg and Cl. In the complex environment of combustion these substances are forming conventional (i.e. SO_x, NO_x) and toxic (PCDD/Fs) pollutants, while, the highly toxic Hg is volatilised in the gas phase mainly as elemental mercury. Aiming to meet the recently adopted strict environmental standards, and the need of affordable in cost clean power production, a preventive fuels pre-treatment technique, based on low temperature carbonization, has been tested. Clean coals were produced from two poor quality Greek coals (Ptolemais and Megalopolis) and an Australian coal sample, in a lab-scale fixed bed reactor under helium atmosphere and ambient pressure. The effect of carbonization temperature (200-900°C) and residence time (5-120 minutes) on the properties of the chars, obtained after pyrolysis, was investigated. Special attention was paid to the removal of pollutants such as S, N, Hg and Cl. To account for possible mineral matter effects, mainly on sulphur removal, tests were also performed with demineralized coal. Reactivity variation of produced clean coals was evaluated by performing non-isothermal combustion tests in a TA Q600 thermogravimetric analyser. Results showed that the low temperature carbonization technique might contribute to clean coal production by effectively removing the major part of the existing polluting compounds contained in coal. Therefore, depending on coal type, nitrogen, mercury and chlorine abatement continuously increases with temperature, while sulphur removal seems to reach a plateau above 500-600°C. Moreover, the prolongation of carbonization time above 20 minutes does not affect the elemental conversion of the pollutants and carbonization at 500-600°C for approx. 20 minutes may be considered sufficient for clean coal production from poor quality coals. Clean coal production at higher pyrolysis temperatures results in observed higher initial combustion temperature, mainly due to lower volatile content in produced chars.

KEYWORDS

[coal](#), [carbonization](#), [clean coal](#), [pollutants removal](#), [reactivity](#)

PAPER SUBMITTED: 2005-11-30

[Authors of this Paper](#)[Related papers](#)[Cited By](#)[External Links](#)

REFERENCES [view full list]

1. Meyers R.A., Coal Structure, Academic Press Inc. New York, 1982
2. Feng J., Li W-Y. Xie K-C. Liu M-R., Li C-Z., Studies of the release rule of NO_x precursors during gasification of coal and its char, Fuel Processing Technology, 84 (2003), pp. 243-254
3. Merdes A., Keener T., Khang S-J., Jenkins R., Investigation into the fate of mercury in bituminous coal during mild pyrolysis, Fuel, 77 (1998), 15, pp. 1783-1792
4. Li W., Lu H., Chen H., Li B., The volatilisation behavior of chlorine in coal during its pyrolysis and CO₂-gasification in a fluidised bed reactor, Fuel 84 (2005), pp. 1874-1878
5. Uzun D., Ozdogan S., Sulfur removal from original and acid treated lignites by pyrolysis, Fuel 85 (2006), pp. 315-322
6. De la Puente G, Fuente E., Pis J.J., Reactivity of pyrolysis chars related to precursor coal chemistry, J. Anal. Appl. Pyrolysis, 53 (2000), pp. 81-93
7. Shaw K., Beamish B., Rodgers K.A., Thermogravimetric analytical procedures for determining reactivities of chars from New Zealand coals, Thermochemica Acta, 302 (1997), pp. 181-187
8. Skodras G., Orfanoudaki T., Kakaras E., Sakellaropoulos G.P., Production of special activated carbon from lignite for environmental purposes, Fuel Processing Technology 77-78 (2002), pp. 75-87.
9. Zabaniotou A.A., Stauroopoulos G., Pyrolysis of used automobile tires and residual char utilization, J. Anal. Appl. Pyrolysis, 70 (2003), pp. 711-722
10. Qi Y., Li W., Chen H., Li B., Sulphur release from coal in fluidised-bed reactor through pyrolysis and partial oxidation with low concentration of oxygen, Fuel 83 (2004), pp. 2189-2194
11. Sentorun C., Haykiri-Acma H., Kucukbayrak S., Effect of mineral matter on the combustion curve of chars, Thermochemica Acta, 277 (1996), pp. 65-73.
12. Haykiri-Acma H., Ersoy-Maricboyu A., Kucukbayrak S., Effect of mineral matter on the reactivity of lignite chars, Energy Conversion & Management, 42 (2001), pp. 11-20.
13. Alonso M. J. G., Borrego A. G., Alvarez D., Parra J. B., Menendez R., Influence of pyrolysis temperature on char optical texture and reactivity, J. Anal. Appl. Pyrolysis, 58-59 (2001), pp. 887-909.
14. Di Blasi C., Buonanno F., Branca C., Reactivities of some biomass chars in air, Carbon, 37 (1999), pp. 1227-1238.
15. Chen G., Yu Q., Sjoström K, Reactivity of char from pyrolysis of birch wood, J Anal. Appl. Pyrolysis, 40-41 (1997), pp. 491-499.
16. Russell N.V., Beelay T.J., Man C. K., Gibbins J. R., Williamson J., Development of TG measurements of intrinsic char combustion reactivity for industrial and research purposes, Fuel Processing Technology, 57 (1998), pp. 113-130.

PDF VERSION [DOWNLOAD]

REMOVAL OF POLLUTANTS FROM POOR QUALITY COALS BY PYROLYSIS

