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N,N'-二(2-羟丙基)哌嗪-H₂SO₄水溶液吸收SO₂动力学

Kinetics of absorption of SO₂ into N,N'-bis(2-hydroxypropyl) piperazine and sulfuric acid aqueous solution

关键词: N N, -二(2-羟丙基)哌嗪;初始速率法;反应动力学;吸收速率

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摘要:在鼓泡式反应器中,以N,N'-二(2-羟丙基)哌嗪(HPP)-硫酸水溶液为吸收剂,利用初始速率法探讨了 SO_2 吸收反应动力学.结果表明:HPP- H_2SO_4 水溶液吸收 SO_2 为快速反应:吸收速率 N_A 随着 SO_2 进口体积浓度 Y_A 和吸收剂浓度 C_N 的增大而增大,随着吸收温度T的升高而降低;吸收速率对 C_N 是0.5级反应,对 Y_A 为0.85级反应.同时,建立了HPP- H_2SO_4 水溶液吸收 SO_2 的吸收速率 N_A 与 C_N 、 Y_A 及T之间的经验关系式,发现该经验关系式的预测值与实验值符合较好,可用于工程设计计算.

Abstract: The aqueous solution of N, N'-bis (2-hydroxypropyl) piperazine (HPP) was used as an absorbent for SO_2 in a bubbling device. The initial rate was used to determine the absorption reaction kinetics of SO_2 . The results showed that the reaction of SO_2 into the aqueous solution of HPP-sulfuric acid is a fast reaction. The absorption rate (N_A) increased with SO_2 concentration in inlet gas (y_A) and the absorbent concentration (C_N) , but decreased with absorption temperature T. The absorption rate was found to be 0.5 and 0.85 order with respect to C_N and y_A , respectively. An empirical correlation of N_A was subsequently developed as a function of C_N , y_A and T, and the absorption rate was calculated. The obtained data were found to be in a good agreement with the experimental absorption rate and hence the empirical correlation could be used for engineering design.

 $\textbf{Key words.} \ \ \underline{N} \ \ \underline{N'} \ \ \text{-bis (2-hydroxypropyl) piperazine; initial rate; reaction kinetics; absorption rate}$

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