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铁锰复合氧化物同时吸附锑镉性能研究

The adsorptive behaviors of Fe-Mn binary oxides towards antimony and cadmium

关键词: 铁锰复合氧化物 五价锑 二价镉 吸附 相互影响

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作者

刘 峰 1. 中国科学院生态环境研究中心 环境水质学国家重点实验室, 北京 100085;

2. 中国科学院大学, 北京 100049

刘锐平 1. 中国科学院生态环境研究中心 环境水质学国家重点实验室, 北京 100085

刘会娟 1. 中国科学院生态环境研究中心 环境水质学国家重点实验室, 北京 100085

曲 久辉 1. 中国科学院生态环境研究中心 环境水质学国家重点实验室, 北京 100085

摘要:以五价锑($\mathsf{Sb}(\mathsf{V})$)和镉(Cd^{2+})为对象,考察了二者单独存在和共存体系下铁锰复合氧化物(FMBO)对其吸附性能,探讨了 $\mathsf{Sb}(\mathsf{V})$ (或 Cd^{2+})的吸附对 Cd²⁺(或Sb(V)) 吸附的影响.研究发现,单独存在体系下Sb(V)和Cd²⁺的吸附常数*K_c*分别为0.48和1.13 L·mg⁻¹,而共存体系下则分别提高至1.88和1.51 L·mg⁻¹; Elovich动力学模型可较好地描述共存条件下Cd²⁺与Sb(V)在FMBO表面的吸附,表明该体系的吸附为多层吸附且为非均相扩散过程;吸附48 h后Sb(V)和Cd²⁺的最大 吸附量分别达到0.32和1.43 mmol·g⁻¹; Sb(V) 在偏酸性而Cd²⁺在偏碱性pH范围具有较好的吸附效果.Sb(V)(或Cd²⁺)通过改变FMBO表面ζ电位和反应平衡pH等机 制影响Cd²⁺(或Sb(V))的吸附.此外,XPS和吸附后水相铁锰浓度结果显示,Cd²⁺可能与FMBO体相中Mn²⁺进行晶格置换并促进Mn²⁺的溶出,进而促进了Sb(V)的吸

Abstract: The adsorption behaviors of Fe-Mn binary oxides (FMBO) towards antimony (Sb(V)) and cadmium (Cd²⁺) in single and co-existed systems were investigated, and the mutual effects between Sb(V) and Cd²⁺ were analyzed. Results indicated that the K_F constants of Sb(V) and Cd²⁺ in Freundlich isotherm were 0.48 and 1.13 L • mg⁻¹ in single system, and increased significantly to 1.88 and 1.51 L • mg⁻¹ in the co-existed system. The Elovich dynamic model can well describe the adsorption of Cd²⁺ and Sb(V) on FMBO in the co-existed system, indicating the multi-layer adsorption with heterogeneous diffusion. After 48 h, the maximum adsorption capacities towards Sb(V) and Cd²⁺ were 0.32 and 1.43 mmol • g⁻¹, respectively. The adsorption of Sb(V) was favored at acidic pH range. By contrast, Cd²⁺ adsorption was favored at alkaline condition. After the adsorption of Cd²⁺ (Sb(V)), the ζ-potential of FMBO and equilibrium pH changed, which affected the adsorption of the Cd²⁺ (Sb(V)). The concentrations of Fe and Mn increased significantly in the presence of Cd²⁺. Accordingly, the elevated ratio of Fe to Mn was confirmed by the XPS analysis. It was concluded that Cd²⁺ facilitated the dissolution of Mn²⁺ from FMBO and therefore increased the adsorption of Sb(V).

Key words: Fe-Mn binary oxide antimony (Sb(V)) cadmium (Cd²⁺) adsorption mutual effects

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