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氮氟掺杂二氧化钛(N,F-TiO₂)的制备及可见光催化活性的研究™

Preparation of N, F-codoped TiO₂ and its photocatalytic activity under visible light

关键词: <u>氦氟掺杂</u> 二氧化钛 制备 可见光 催化活性

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基金项目: 辽宁省科技厅自然科学基金项目(No.201202091);辽宁省教育厅自然科学基金项目(No.L2012006);辽宁省教育厅特色学科建设项目;辽宁省水环境生物监测与水生态安全重点实验室项目;辽宁大学"211"工程建设项目(No.HJ211007);辽宁环境科研教育"123工程"(No.CEPF2009-123-2-10)

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摘要:以氨水为氮源,氢氟酸为氟源,采用溶胶-凝胶法制备了氮氟掺杂二氧化钛(N,F-TiO₂)光催化剂,并通过X-射线衍射(XRD)和扫描电镜(SEM)技术对其晶型和形态进行表征.最后以酸性红B为模型污染物,探讨了N和F加入量、焙烧温度、焙烧时间、照射时间、催化剂用量、溶液初始浓度和照射功率等因素对N,F-TiO₂可见光催化活性的影响.结果表明,制备的N,F-TiO₂以锐钛型为主,N和F的掺杂对TiO₂的晶相没有明显改变,但可以扩大TiO₂的可见光响应范围.当N和F的加入量均为2.0%,且在500 ℃下焙烧40 min时,得到的N,F-TiO₂的可见光催化活性明显高于单N或单F掺杂的TiO₂(N-TiO₂或F-TiO₂).对于50 mL浓度为10.0 mg · L⁻¹的酸性红B溶液,当催化剂加入量为1.5 g · L⁻¹,128 W光照3.0 h,溶液pH=5.6时,去除率为85.40%.适当延长光照时间至4.0 h,降解率几乎可达100%.另外,研究还证明了N,F-TiO₂催化可见光降解过程中有 · OH自由基生成.

Abstract: Preparation of N,F-codoped TiO₂ (N,F-TiO₂) photocatalyst using ammonia water as nitrogen sources and hydrofluoric acid as fluorine sources by sol-gel method was proposed. The crystal form and morphology of the as-prepared photocatalysts were characterized by X-ray diffraction (XRD) and scanning electron microscope (SEM). The influence of N-to-Ti and F-to-Ti molar ratio, calcination temperature and time, irradiation time, catalyst dose, initial concentration of solution and light power on catalytic activity of N,F-TiO₂ under visible light was investigated using acid red B as a model contaminant. The results showed that anatase is the main crystal form of N,F-TiO₂. Also, the doping of nitrogen and fluorine produced no significant change in the crystalline phase of TiO₂, but it could expand the range of response of TiO₂ to visible light. The highest catalytic activity of N,F-TiO₂ could be obtained at 2.0% molar ratio of N,F/Ti while calcining at 500 °C for 40 min, and it was obviously higher than that of N-TiO₂ or F-TiO₂. The degradation ratio reached 85.40% for 50 mL solution with 10 mg • L⁻¹ acid red B at 1.5 g • L⁻¹ N,F-TiO₂ dose, 3.0 h light irradiation, 128 W light power and pH=5.6. The removal ratio could reach nearly 100% by increasing the irradiation time to 4.0 h with no change of other conditions. In addition, the hydroxyl radicals (• OH) were proved to be generated in the catalyzed degradation by N,F-TiO₂ under visible light.

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