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纳米Fe^O对Cr(VI)的还原及其影响因素

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Reduction of Hexavalent Chromium by Nanoscale Fe⁰ and Its Influencing Factors

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摘要

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Supporting Info

摘要 采用纳米 Fe^0 还原水溶液中的Cr(VI),考察纳米 Fe^0 投加量、Cr(VI)初始浓度、溶液pH值和有机酸等因素对Cr(VI)还原的影响。结果表明,纳米 Fe^0 对Cr(VI)的还原效果明显,其对Cr(VI)的还原率分别是铁粉和铁屑的 $Tempsize{1}$ 和13倍。Cr(VI) 溶液初始质量浓度为 $Tempsize{2}$ 20 mg· $Tempsize{1}$ 1、 $Tempsize{2}$ 20 mg· $Tempsize{2}$ 3。容液低 $Tempsize{2}$ 4 m时纳米 $Tempsize{2}$ 6 mg· $Tempsize{2}$ 7 mg· $Tempsize{2}$ 7 mg· $Tempsize{2}$ 8 mg· $Tempsize{2}$ 8 mg· $Tempsize{2}$ 8 mg· $Tempsize{2}$ 9 mg· $Tempsize{2$

关键词: 纳米Fe^O Cr(VI) 还原 影响因素

Abstract: An experiment was carried out on nanoscale zero-valent iron (NZVI) reducing Cr(VI) in water to explore effects of NZVI application rate, initial concentrations of Cr(VI), initial pHs, and organic acids on the reduction. It was found that NZVI efficiently reduced Cr(VI) in water at a rate 7 and 13 times higher than that of iron powder and fillings, respectively. Under the condition of the initial concentration of Cr(VI) being 20 $mg \cdot L^{-1}$ and the NZVI application rate being $5g \cdot L^{-1}$, the reduction rate reached 82% after 24 h of incubation. Low pH solution pH promoted corrosion of NZVI, which raised Cr(VI) reduction rate. The highest reduction rate occurred in solution being 3.0 in pH. Among organic acids, oxalic acid, malonic acid and succinic acid all significantly improved the effect of NZVI reducing Cr(VI), and followed the order of oxalic acid > malonic acid > succinic acid.

Keywords: nanoscale zero-valent iron (NZVI) hexavalent chromium reduction influencing factor

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