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### Feasibility study of decomposing methane with hydroxyl radicals

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#### Abstract

Coal-mine gas disaster is one of the most serious coal-mine disasters in China. The main component of coal-mine gas, methane is chemically stable and very difficult to be degraded by conventional methods. Hydroxyl radical ( $\cdot\text{OH}$ ), due to strong oxidizing ability and high electro-negativity, is the primary degradation source of atmospheric methane. In the present study, methane degradation using hydroxyl radicals generated by Fenton's reagent,  $\text{Fe}^{2+}/\text{H}_2\text{O}_2$ , has been carried out in the self-designed bubbling reactor. The effects of  $\text{H}_2\text{O}_2$  concentration, dosage of  $\text{FeSO}_4 \cdot 7\text{H}_2\text{O}$  and initial pH value on methane removal efficiency were investigated respectively. It has been found that the optimal reaction conditions were 100 mM of hydrogen peroxide, 2.00 mM of ferrous ion and initial pH value of 2.5. Under optimal conditions, the removal efficiency of methane reached 25% after 30 min. The preliminary experimental results unambiguously demonstrate that the degradation of methane using hydroxyl radicals generated by Fenton's reagent is feasible.

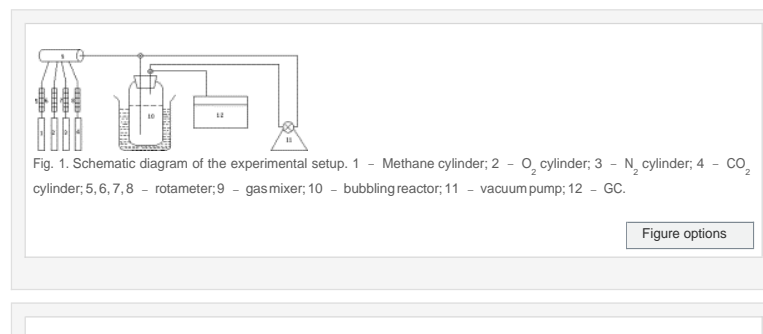
#### Highlights

► In coal-mine gas, methane is stable and difficult to be degraded by common methods. ► Methane degradation using hydroxyl radicals was conducted in the bubbling reactor. ► The primary factors, which would influence the methane removal, were investigated. ► The removal efficiency of methane reached 25% under the optimal conditions. ► The results show methane degradation *via* hydroxyl radical oxidation is feasible.

#### Keywords

Methane; Hydroxyl radical; Degradation; Fenton's reagent

#### Figures and tables from this article:


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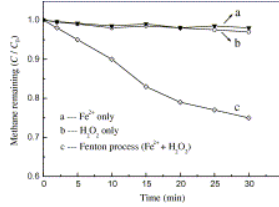


Fig. 2. Methane removal efficiency by various methods. Experimental conditions: 1.50 mM Fe<sup>2+</sup>, 100 mM H<sub>2</sub>O<sub>2</sub>, pH 2.5, T = 25 ° C.

Figure options

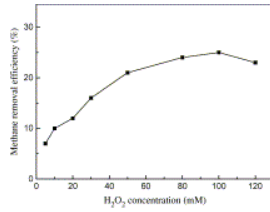


Fig. 3. Effects of H<sub>2</sub>O<sub>2</sub> concentration on methane removal efficiency by hydroxyl radical oxidation via Fenton reagent. Experimental conditions: 2.00 mM Fe<sup>2+</sup>, pH 2.5, T = 25 ° C.

Figure options

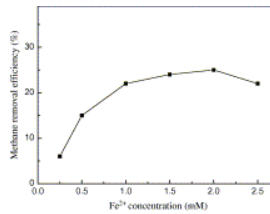


Fig. 4. Effects of ferrous ion concentration on methane removal efficiency by hydroxyl radical oxidation through Fenton reaction. Experimental conditions: 100 mM H<sub>2</sub>O<sub>2</sub>, pH 2.5, T = 25 ° C.

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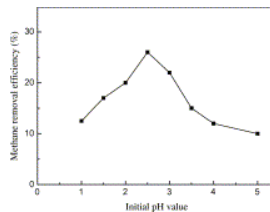


Fig. 5. Effects of initial pH value on methane degradation efficiency by hydroxyl radical oxidation via Fenton process. Experimental conditions: 2.00 mM Fe<sup>2+</sup>, 80 mM H<sub>2</sub>O<sub>2</sub>, T = 25 ° C.

Figure options



The First International Symposium on Mine Safety Science and Engineering (ISMSE2011) will be held in Beijing on October 26 – 29, 2011. The symposium is authorized by the State Administration of Work Safety and is sponsored by China Academy of Safety Science & Technology (CASST), China University of Mining & Technology (Beijing) (CUMTB), Datong Coal Mine Group, McGill University (Canada) and University of Wollongong (Australia) with participation from several other universities from round the world, research institutes, professional associations and large enterprises. The topics will focus on mines safety field: theory on mine safety science and engineering technology, coal mine safety science & engineering technology, metal and nonmetal mines safety science & engineering technology, petroleum and natural gas exploitation safety science & engineering technology, mine safety management and safety standardization science & technology, occupational health and safety in mine, emergent rescue engineering technology in mine, etc.



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