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Sorption of heavy metals in organic horizons of acid forest soils at low added concentrations

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The individual adsorption/desorption of Cu, Pb, and Zn in six soil samples taken from organic horizons of acid forest soils is studied in batch experiments. These three metals were chosen because of the varying extent of their sorption to the soil organic matter. The initial concentration range for all metals is 0–200 mg/l. 0.01 mol/l Ca(NO₃)₂ was used as the background electrolyte. Soil samples were taken at random locations throughout a relatively small area and are characterized by small differences in mineralogical composition. Organic matter content and pH of soils are considered as the key factors causing differences in sorption/desorption behaviour of selected metals. In general, the sorption of all the three metals increased with a combination of higher pH value and increasing organic matter content. Anyway, the extent of the influence of these two characteristics on sorption behaviour of metals is different, for desorption significant differences were not determined. Copper sorption seems to reflect mainly the differences in organic matter content in addition to the differences in pH. The average amount of sorbed Pb was approximately 90%, Cu sorption averaged 60%, and that of Zn 30%. Desorption of Pb into 0.01 mol/l Ca(NO₃)₂ remained at approximately 4%, for Zn at 30%, and desorption of Cu reached up to 13% of the amount adsorbed. All the studied soils proved effective at immobilizing lead and copper, but zinc was relatively highly released from these soils, even when the sorbed amount was minimal. Different vegetational background of these samples (either beech or spruce forests) does not significantly influence the sorption extent of these metals as the lower pH of samples taken under spruce stands is probably compensated by higher organic matter content in these samples. The experimental data are fitted by the Freundlich equation and the parameters of this equation together with the adsorption and desorption efficiency are used for comparison of the behaviour of all the three metals among the six soils.

Keywords:

acidic soils; adsorption; desorption; organic matter; risk elements

References:

Abate Gilberto, Masini Jorge C. (2002): Complexation of Cd(II) and Pb(II) with humic acids studied by anodic stripping voltammetry using differential equilibrium functions and discrete site models. *Organic Geochemistry*, 33, 1171-1182 [https://doi.org/10.1016/S0146-6380\(02\)00087-6](https://doi.org/10.1016/S0146-6380(02)00087-6)

Adhikari Tapan, Singh M.V (2003): Sorption characteristics of lead and cadmium in some soils of India. *Geoderma*, 114, 81-92 [https://doi.org/10.1016/S0016-7061\(02\)00352-X](https://doi.org/10.1016/S0016-7061(02)00352-X)

Adriano D.C. (2001): Trace Elements in Terrestrial Environments: Biogeochemistry, Bioavailability, and Risks of Metals. New York/Berlin/Heidelberg, Springer-Verlag.

Alloway B.J. (1995): Soil processes and the behaviour of heavy metals. In: Alloway B.J. (ed.): *Heavy Metals in Soils*. London, Blackie Academic and Professional: 11–37.

Arias M., Pérez-Novo C., López E., Soto B. (2006): Competitive adsorption and desorption of copper and zinc in acid soils. *Geoderma*, 133, 151-159 <https://doi.org/10.1016/j.geoderma.2005.07.002>

Bäckström Mattias, Dario Mårten, Karlsson Stefan, Allard Bert (2003): Effects of a fulvic acid on the adsorption of mercury and cadmium on goethite. *Science of The Total Environment*,

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Bolster Carl H. (2008): Revisiting a Statistical Shortcoming when Fitting the Langmuir Model to Sorption Data. *Journal of Environment Quality*, 37, 1986-
<https://doi.org/10.2134/jeq2007.0461>

Bradl Heike B (2004): Adsorption of heavy metal ions on soils and soils constituents. *Journal of Colloid and Interface Science*, 277, 1-18 <https://doi.org/10.1016/j.jcis.2004.04.005>

Christl Iso, Milne Chris J., Kinniburgh David G., Kretzschmar Ruben (2001): Relating Ion Binding by Fulvic and Humic Acids to Chemical Composition and Molecular Size. 2. Metal Binding. *Environmental Science & Technology*, 35, 2512-2517
<https://doi.org/10.1021/es0002520>

Elliott H. A., Liberati M. R., Huang C. P. (1986): Competitive Adsorption of Heavy Metals by Soils. *Journal of Environment Quality*, 15, 214-
<https://doi.org/10.2134/jeq1986.00472425001500030002x>

Escrig I., Morell I. (1997): Effect of calcium on the soil adsorption of cadmium and zinc in some Spanish sandy soils. *Water, Air, and Soil Pollution*, 105: 507–520.

Gerritse R. G., van Driel W. (1984): The Relationship Between Adsorption of Trace Metals, Organic Matter, and pH in Temperate Soils. *Journal of Environment Quality*, 13, 197-
<https://doi.org/10.2134/jeq1984.00472425001300020005x>

Gondar D., López R., Fiol S., Antelo J.M., Arce F. (2006): Cadmium, lead, and copper binding to humic acid and fulvic acid extracted from an ombrotrophic peat bog. *Geoderma*, 135, 196-203 <https://doi.org/10.1016/j.geoderma.2005.12.003>

GREGOR J. E., POWELL H. K. J., TOWN R. M. (1989): Evidence for aliphatic mixed mode coordination in copper(II)-fulvic acid complexes. *Journal of Soil Science*, 40, 661-673
<https://doi.org/10.1111/j.1365-2389.1989.tb01307.x>

Hooda P.S., Alloway B.J. (1998): Cadmium and lead sorption behaviour of selected English and Indian soils. *Geoderma*, 84, 121-134 [https://doi.org/10.1016/S0016-7061\(97\)00124-9](https://doi.org/10.1016/S0016-7061(97)00124-9)

Javorský P. (1987): *Chemical Analysis in Agricultural Laboratories*. Praha, MZe. (in Czech)

Karpukhin A. I., Bushuev N. N. (2007): Distribution of heavy metals by the molecular-weight fractions of humic acids in the soils of long-term field experiments. *Eurasian Soil Science*, 40, 265-273 <https://doi.org/10.1134/S1064229307030040>

Lair G. J., Gerzabek M. H., Haberhauer G. (2007): Sorption of heavy metals on organic and inorganic soil constituents. *Environmental Chemistry Letters*, 5, 23-27
<https://doi.org/10.1007/s10311-006-0059-9>

MURRAY K., LINDER P. W. (1984): Fulvic acids: structure and metal binding. *Journal of Soil Science*, 35, 217-222 <https://doi.org/10.1111/j.1365-2389.1984.tb00277.x>

Nelson D.W., Sommers L.E. (1996): Total carbon, organic carbon, and organic matter. In: Page A.L. (ed.): *Methods of Soil Analysis. Part 2. Chemical and Microbiological Properties*. Madison, American Society of Agronomy, Inc., Soil Science Society of America, Inc.: 961–1010.

OECD (2001): *Guidelines for the testing of chemicals/Section 1: Physical-chemical properties test No. 106: Adsorption – desorption using a batch equilibrium method*. Adopted 21.1.2000.

Pavů L., Borůvka L., Nikodem A., Rohošková M., Penížek V. (2007): Altitude and forest type effects on soils in the Jizera Mountains region. *Soil and Water Resources*, 2: 35–44.

Sastre Jordi, Rauret Gemma, Vidal Miquel (2007): Sorption–desorption tests to assess the risk derived from metal contamination in mineral and organic soils. *Environment International*, 33, 246-256 <https://doi.org/10.1016/j.envint.2006.09.017>

SCHNITZER M., SKINNER S. I. M. (1967): ORGANO-METALLIC INTERACTIONS IN SOILS. *Soil Science*, 103, 247-252 <https://doi.org/10.1097/00010694-196704000-00004>

Senesi Nicola, Sposito Garrison, Holtzclaw Kenneth M., Bradford Gordon R. (1989): Chemical Properties of Metal-Humic Acid Fractions of a Sewage Sludge-Amended Aridisol. *Journal of Environment Quality*, 18, 186- <https://doi.org/10.2134/jeq1989.00472425001800020010x>

Skinner M. F., Zabowski D., Harrison R., Lowe A., Xue D. (): Measuring the cation exchange capacity of forest soils. *Communications in Soil Science and Plant Analysis*, 32, 1751-1764 <https://doi.org/10.1081/CSS-120000247>

Suchara I., Sucharová J. (2002): Distribution of sulphur and heavy metals in forest floor humus of the Czech Republic. *Water, Air and Soil Pollution*, 136: 289–316.

Tejnecký Václav, Drábek Ondřej, Borůvka Luboš, Nikodem Antonín, Kopáč Jan, Vokurková Petra, Šebek Ondřej (2010): Seasonal variation of water extractable aluminium forms in acidified forest organic soils under different vegetation cover. *Biogeochemistry*, 101, 151-163 <https://doi.org/10.1007/s10533-010-9450-5>

Tejnecký Václav, Bradová Monika, Borůvka Luboš, Němeček Karel, Šebek Ondřej, Nikodem Antonín, Zenáhlíková Jitka, Rejzek Jan, Drábek Ondřej (2013): Profile distribution and temporal changes of sulphate and nitrate contents and related soil properties under beech and spruce forests. *Science of The Total Environment*, 442, 165-171 <https://doi.org/10.1016/j.scitotenv.2012.10.053>

Tipping E (1981): The adsorption of aquatic humic substances by iron oxides. *Geochimica et Cosmochimica Acta*, 45, 191-199 [https://doi.org/10.1016/0016-7037\(81\)90162-9](https://doi.org/10.1016/0016-7037(81)90162-9)

Trakal L., Komárek M., Száková J., Tlustoš P., Tejnecký V., Drábek O. (2012): Sorption Behavior of Cd, Cu, Pb, and Zn and Their Interactions in Phytoremediated Soil. *International Journal of Phytoremediation*, 14, 806-819 <https://doi.org/10.1080/15226514.2011.628714>

Violante A, Cozzolino V, Perelomov L, Caporale A.G, Pigna M (2010): MOBILITY AND BIOAVAILABILITY OF HEAVY METALS AND METALLOIDS IN SOIL ENVIRONMENTS. *Journal of soil science and plant nutrition*, 10, - <https://doi.org/10.4067/S0718-95162010000100005>

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