

论文

H₂O₂存在下橙皮苷的极谱催化波机理及其应用

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摘要:

目的 研究在过氧化氢存在下橙皮苷极谱催化波产生机理,建立测定橙皮苷的极谱催化波新方法。方法用线性变位极谱法、循环伏安法等技术。结果 橙皮苷C-4位上的羰基C=O首先经单电子单质子还原为自由基,产生第1个还原波;该自由基的一部分由于共轭作用使其能量降低,并进一步还原,产生第2个还原波,另一部分发生二聚化反应。当氧化剂H₂O₂存在时,H₂O₂氧化橙皮苷羰基还原中间体自由基,阻断了该自由基进一步还原和二聚化反应,并使橙皮苷再生,产生橙皮苷的极谱催化波。在0.12mol·L⁻¹ HAc-0.40mol·L⁻¹ NaAc (pH 5.3)-1.0×10⁻²mol·L⁻¹ H₂O₂支持电解质中,该催化波的一阶导数

关键词: 橙皮苷 过氧化氢 极谱催化波 自由基

MECHANISM OF THE POLAROGRAPHIC CATALYTIC WAVE OF HESPERIDIN IN THE PRESENCE OF HYDROGEN PEROXIDE AND ITS APPLICATION

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Abstract:

AIM To study the mechanism of a polarographic catalytic wave of hesperidin in the presence of hydrogen peroxide, and to propose a new catalytic wave method for the determination of hesperidin. METHODS Linear-potential scan polarography, cyclic voltammetry and DC polarography were used in this work. RESULTS In 0.12 mol·L⁻¹ HAc-0.40 mol·L⁻¹ NaAc (pH 5.3) supporting electrolyte, the carbonyl group C=O of hesperidin at the C-4 position yielded two reduction waves. The carbonyl group C=O firstly undergoes a 1 e⁻, 1H⁺ reduction to form an intermediate free radical, producing the first reduction wave with peak potential -1.35 V (vs SCE). Further reduction of the radical was simultaneous with its dimerization. A part of the radical was further reduced, producing the second reduction wave with peak potential -1.70 V (vs SCE). Other part of the free radical was dimerized. The dimer could be oxidized at -1.17 V (vs SCE). When oxidant H₂O₂ was present, both the further reduction and dimerization were interrupted, and a polarographic catalytic wave was produced because H₂O₂ oxidized the intermediate free radical to regenerate the original C=O group. Based on the polarographic catalytic wave, a new method for the determination of hesperidin was proposed. 0.12 mol·L⁻¹ HAc-0.40 mol·L⁻¹ NaAc (pH 5.3)-1.0×10⁻² mol·L⁻¹ H₂O₂ solution was chosen as supporting electrolyte. The first-order derivative technique was used to eliminate the higher background current from H₂O₂ reduction and to well separate the catalytic wave from H₂O₂ reduction wave. The first-order derivative peak current of the catalytic wave is proportional to hesperidin concentration in the range of 1.0×10⁻⁷ -1.8×10⁻⁶ mol·L⁻¹ with

Keywords: hydrogen peroxide polarographic catalytic wave free radical hesperidin

收稿日期 2001-03-09 修回日期 网络版发布日期

DOI:

基金项目:

通讯作者:

作者简介:

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