

ONLINE ISSN : 1881-1361 PRINT ISSN : 0287-4547

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Dental Materials Journal

Vol. 27 (2008), No. 6 p.765-774

[PDF (795K)] [References]

UV-VIS spectra and photoinitiation behaviors of acylphosphine oxide and bisacylphosphine oxide derivatives in unfilled, light-cured dental resins

<u>Kunio IKEMURA¹</u>, <u>Kensuke ICHIZAWA¹</u>, <u>Mariko YOSHIDA¹</u>, <u>So ITO¹</u> and Takeshi ENDO²⁾³⁾

1) Department of Research and Development, Shofu Inc.

2) Molecular Engineering Institute, Kinki University

3) Chemical Resources Laboratory, Tokyo Institue of Technology

(Received February 7, 2008) (Accepted May 16, 2008)

Abstract:

The aims of this study were to investigate the photoinitiation behaviors of acylphosphine oxide (APO) and bisacylphosphine oxide (BAPO) derivatives in comparison to D,L-camphorquinone (CQ)/tertiary amine (EDAB) system. Fifty six kinds of experimental unfilled, light-cured resins were prepared using APO, BAPO, CQ, EDAB, resin monomers, and adhesive monomers. The measurements of ultraviolet-visible (UV-VIS) spectroscopy, photopolymerization with differential scanning calorimeter (photo-DSC) and degree of conversion (DC) determination were performed. Results showed that the UV-VIS spectra of all APO and BAPO derivatives possessed λ max ranging between 365 and 416 nm. Their photoinitiation behaviors were not influenced by acidic adhesive monomers formulated in unfilled Bis-GMA-based resins (p<0.05). Although BAPO exhibited higher reactivity than CQ/EDAB in unfilled 6-methacryloyloxyhexcyl phosphonoacetate (6-MHPA)/ 2-hydroxyethyl methacrylate (HEMA) and 6-MHPA/glycerol monomethacrylate (GM)-based resins, it was found that there were no statistically significant differences in DC (%) between BAPO [44.2(6.5)–51.2(4.3)] and CQ/EDAB [42.4(4.4)–47.5(3.7)] (p<0.05). It was concluded that APO and BAPO photoinitiators exhibited reactivity comparable to that of

CQ/tertiary amine system.

Key words: <u>UV-VIS spectra</u>, <u>Photo-DSC</u>, <u>Degree of conversion</u>



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To cite this article:

Kunio IKEMURA, Kensuke ICHIZAWA, Mariko YOSHIDA, So ITO and Takeshi ENDO. UV-VIS spectra and photoinitiation behaviors of acylphosphine oxide and bisacylphosphine oxide derivatives in unfilled, light-cured dental resins . Dent. Mater. J. 2008; 27: 765-774 .

doi:10.4012/dmj.27.765 JOI JST.JSTAGE/dmj/27.765

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